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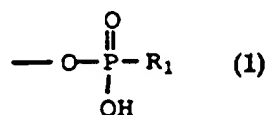
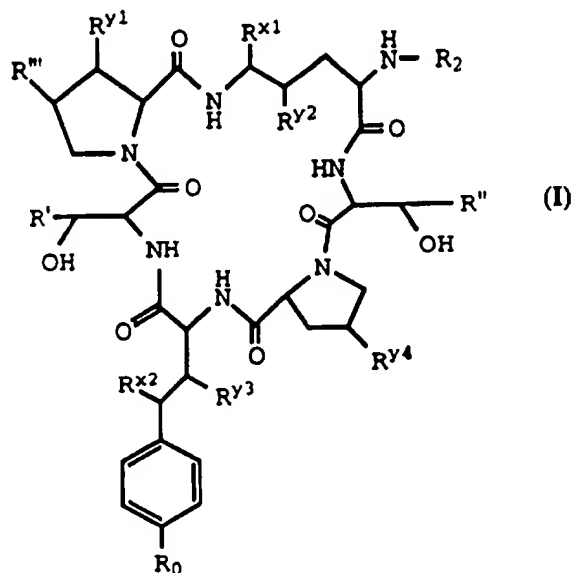
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(54) Title: CYCLIC PEPTIDE ANTIFUNGAL AGENTS

(57) Abstract

Provided are pharmaceutical formulations, and methods of inhibiting fungal and parasitic activity using a compound of formula (I) wherein: R' is hydrogen, methyl or NH<sub>2</sub>C(O)CH<sub>2</sub>; R'' and R''' are independently methyl or hydrogen; R<sup>x1</sup>, R<sup>x2</sup>, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup>, and R<sup>y4</sup> are independently hydroxy or hydrogen; R<sub>0</sub> is a group of formula (1); R<sub>1</sub> is C<sub>1</sub>-C<sub>6</sub> alkyl, C<sub>1</sub>-C<sub>6</sub> alkoxy, phenyl, p-halo-phenyl, p-nitrophenyl, phenoxy, benzyl, p-halo-benzyl, or p-nitro-benzyl; and R<sub>2</sub> is an acyl side chain as defined herein.



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## CYCLIC PEPTIDE ANTIFUNGAL AGENTS

5 This invention relates to semi-synthetic cyclic peptide compounds which are useful as antifungal and antiparasitic agents and which have improved stability and water solubility. In particular, it relates to derivatives of the echinocandin class of cyclic peptides; to methods for treating fungal and parasitic infections, and to formulations useful in the methods.

10 The compounds provided by this invention are semi-synthetic compounds derived from cyclic peptides which are produced by culturing various microorganisms. A number of cyclic peptides are known in the art including echinocandin B (A30912A), aculeacin, mulundocandin, sporiofungin, L-671,329, and S31794/F1.

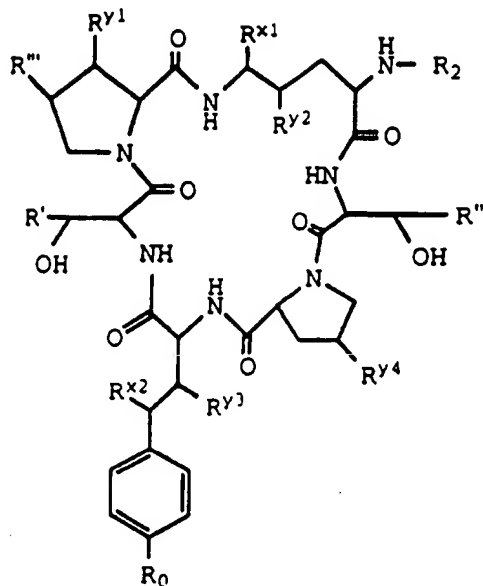
15 In general, these cyclic peptides may be structurally characterized as a cyclic hexapeptide core (or nucleus) with an acylated amino group on one of the core amino acids. The amino group is typically acylated with a fatty acid group forming a side chain off the nucleus. For example, echinocandin B has a linoleoyl side chain while aculeacin has a palmitoyl side chain.

20 The fatty acid side chains may be removed from the cyclic peptide core to provide an amino nucleus (for example, a compound of formula I, below, where R<sub>2</sub> is hydrogen). The amino group may then be re-acylated to provide semi-synthetic compounds such as those claimed in the present application.

25 The echinocandin B nucleus has been re-acylated with certain non-naturally occurring side chain moieties to provide a number of antifungal agents (see, Debono, U.S. Pat. No. 4,293,489). Among such antifungal agents is cilofungin which is represented by a compound of formula I where R', R'', and R''' are methyl; R<sup>x1</sup> and R<sup>x2</sup> are hydroxy, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup>, R<sup>y4</sup> are hydroxy, R<sub>0</sub> is hydroxy, and R<sub>2</sub> is p-(octyloxy)benzoyl.

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The present invention provides a compound of formula I:



I

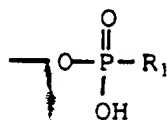
5 wherein:

R' is hydrogen, methyl or  $\text{NH}_2\text{C}(\text{O})\text{CH}_2-$ ;

R'' and R''' are independently methyl or hydrogen;

$\text{R}^{\text{x}1}$ ,  $\text{R}^{\text{x}2}$ ,  $\text{R}^{\text{y}1}$ ,  $\text{R}^{\text{y}2}$ ,  $\text{R}^{\text{y}3}$ , and  $\text{R}^{\text{y}4}$  are independently hydroxy or hydrogen;

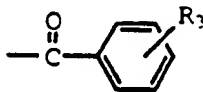
10  $\text{R}_0$  is a group of the formula



$\text{R}_1$  is  $\text{C}_1$ - $\text{C}_6$  alkyl,  $\text{C}_1$ - $\text{C}_6$  alkoxy, phenyl, p-halo-phenyl, p-nitrophenyl, phenoxy, benzyl, p-halo-benzyl, or p-nitro-benzyl;

15

I)  $\text{R}_2$  is a group of the formula



20

where:

A)  $\text{R}_3$  is  $\text{C}_1$ - $\text{C}_{12}$  alkyl,  $\text{C}_1$ - $\text{C}_6$  alkoxy or quinolyl;

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B)  $R_3$  is  $-O-(CH_2)_m-[O-(CH_2)_n]_p-O-(C_1-C_{12} \text{ alkyl})$ ;

$m$  and  $n$  are independently 2, 3 or 4;

$p$  is 0 or 1; or

C)  $R_3$  is  $-Y-(C_1-C_{12} \text{ alkyl})$ ;

5  $Y$  is  $-C\equiv C-$  or  $-CH=CH-$ ; or

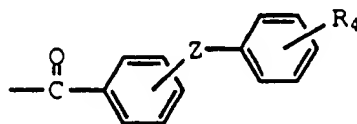
D)  $R_3$  is  $-O-(CH_2)_q-G$ ;

$q$  is 2, 3 or 4;

$G$  is  $C_7-C_{10}$  bicycloalkyl or  $C_7-C_{14}$  tricycloalkyl; or

10

II)  $R_2$  is a group of the formula



where:

15  $Z$  is  $-O-$ ,  $-C\equiv C-$ ,  $-CH=CH-$ ,  $-CH_2-CH_2-$ ,  $-CH_2-$ , or a bond;

A)  $R_4$  is hydrogen,  $C_1-C_{12}$  alkyl,  $C_1-C_{12}$  substituted alkyl,  $C_2-C_{12}$  alkenyl,  $C_2-C_{12}$  substituted alkenyl,  $C_2-C_{12}$  alkynyl,  $C_2-C_{12}$  substituted alkynyl,  $C_1-C_{12}$  alkoxy,  $C_3-C_{12}$  cycloalkyl,  $C_7-C_{10}$  bicycloalkyl,  $C_7-C_{14}$  tricycloalkyl,  $C_3-C_{12}$  cycloalkoxy, naphthyl, pyridyl, thienyl, benzothienyl, quinolyl or phenyl; or

20

B)  $R_4$  is phenyl substituted by amino,  $C_1-C_{12}$  alkylthio, halo,  $C_1-C_{12}$  alkyl,  $C_2-C_{12}$  alkenyl,  $C_2-C_{12}$  alkynyl,  $C_1-C_{12}$  substituted alkyl,  $C_2-C_{12}$  substituted alkenyl,  $C_2-C_{12}$  substituted alkynyl,  $C_1-C_{12}$  alkoxy, trifluoromethyl, phenyl, substituted phenyl, or phenyl substituted with a group of the formula  $-O-(CH_2)_m-[O-(CH_2)_n]_p-O-(C_1-C_{12} \text{ alkyl})$  where  $m$ ,  $n$  and  $p$  are as defined above; or

25

30

C)  $R_4$  is  $C_1-C_{12}$  alkoxy substituted with halo,  $C_3-C_{12}$  cycloalkyl,  $C_7-C_{10}$  bicycloalkyl,  $C_7-C_{14}$  tricycloalkyl,  $C_1-C_6$  alkoxy,  $C_2-C_{12}$  alkynyl, amino,  $C_1-C_4$  alkylamino,

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di(C<sub>1</sub>-C<sub>4</sub> alkyl)amino, formamido, C<sub>2</sub>-C<sub>12</sub> alkanoylamino, or phenyl substituted with a group of the formula -O-(CH<sub>2</sub>)<sub>m</sub>-[O-(CH<sub>2</sub>)<sub>n</sub>]<sub>p</sub>-O-(C<sub>1</sub>-C<sub>12</sub> alkyl) where m, n and p are as defined above; or

5

D) R<sub>4</sub> is -O-(CH<sub>2</sub>)<sub>r</sub>-W-R<sub>5</sub>;

r is 2, 3 or 4;

W is pyrrolidino, piperidino or piperazino;

10 R<sub>5</sub> is hydrogen, C<sub>1</sub>-C<sub>12</sub> alkyl, C<sub>3</sub>-C<sub>12</sub> cycloalkyl, benzyl or C<sub>3</sub>-C<sub>12</sub> cycloalkylmethyl; or

E) R<sub>4</sub> is -Y<sup>1</sup>-R<sub>6</sub>;

Y<sup>1</sup> is -C≡C- or -CH=CH-;

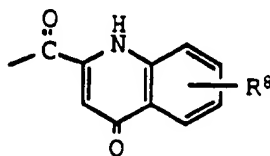
15 R<sub>6</sub> is C<sub>3</sub>-C<sub>12</sub> cycloalkyl, C<sub>7</sub>-C<sub>10</sub> bicycloalkyl, C<sub>7</sub>-C<sub>14</sub> tricycloalkyl, C<sub>3</sub>-C<sub>12</sub> cycloalkenyl, naphthyl, benzothiazolyl, thienyl, indanyl, fluorenyl, or phenyl substituted with C<sub>1</sub>-C<sub>12</sub> alkylthio, C<sub>2</sub>-C<sub>12</sub> alkenyl, C<sub>2</sub>-C<sub>12</sub> alkynyl, halo(C<sub>1</sub>-C<sub>6</sub> alkoxy) or a group of the formula -O-(CH<sub>2</sub>)<sub>r</sub>-W-R<sub>5</sub> where r, W and R<sub>5</sub> are as defined above; or

20 R<sub>6</sub> is phenyl substituted with a group of the formula -O-(CH<sub>2</sub>)<sub>m</sub>-[O-(CH<sub>2</sub>)<sub>n</sub>]<sub>p</sub>-O-(C<sub>1</sub>-C<sub>12</sub> alkyl) where m, n and p are as defined above; or

25 F) R<sub>4</sub> is C<sub>1</sub>-C<sub>12</sub> alkoxy substituted with a group of the formula -NHC(O)R<sub>7</sub>;

R<sub>7</sub> is C<sub>1</sub>-C<sub>6</sub> alkoxy, or phenyl(C<sub>1</sub>-C<sub>6</sub> alkoxy); or

III) R<sub>2</sub> is a group of the formula

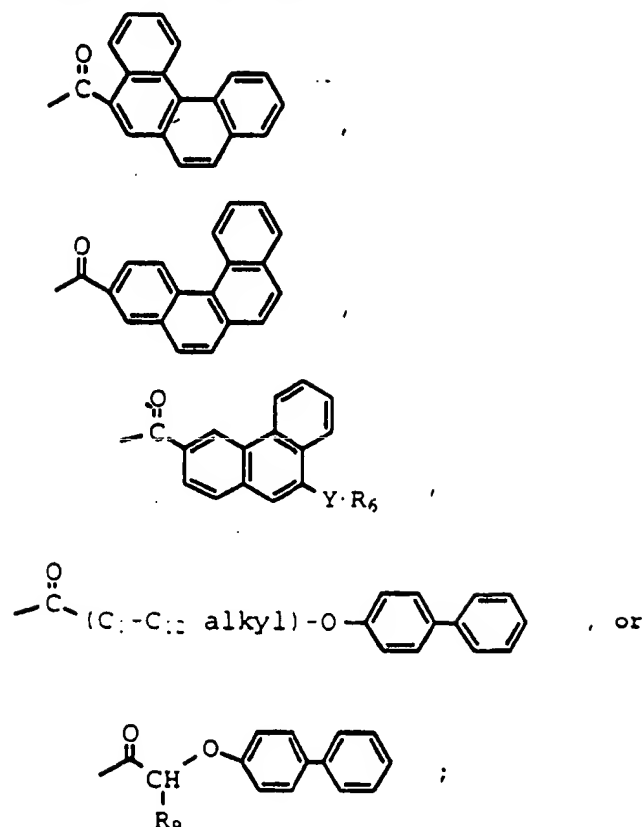


30

where R<sup>8</sup> is C<sub>1</sub>-C<sub>12</sub> alkoxy or a group of the formula -O-(CH<sub>2</sub>)<sub>m</sub>-[O-(CH<sub>2</sub>)<sub>n</sub>]<sub>p</sub>-O-(C<sub>1</sub>-C<sub>12</sub> alkyl) where m, n and p are as defined above; or

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IV)  $R_2$  is a group of the formula



where:

5

$Y$  and  $R_6$  are as defined above;

$R_9$  is phenyl,  $C_1$ - $C_{12}$  alkyl, or  $C_1$ - $C_{12}$  alkoxy; or

V)  $R_2$  is naphthoyl substituted with  $R_4$  where  $R_4$  is as defined above;

10 or a pharmaceutically acceptable salt thereof.

Also provided are pharmaceutical formulations, methods for inhibiting parasitic or fungal activity and methods of treating fungal or parasitic infections which employ the compounds of the invention.

15

As used herein, the term " $C_1$ - $C_{12}$  alkyl" refers to a straight or branched alkyl chain having from one to twelve carbon atoms. Typical  $C_1$ - $C_{12}$  alkyl groups include methyl, ethyl, propyl, isopropyl, butyl, sec-butyl, t-butyl, pentyl, 5-methylpentyl, hexyl, heptyl, 3,3-

20

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dimethylheptyl, octyl, 2-methyl-octyl, nonyl, decyl, undecyl, dodecyl and the like. The term "C<sub>1</sub>-C<sub>12</sub> alkyl" includes within its definition the terms "C<sub>1</sub>-C<sub>6</sub> alkyl" and C<sub>1</sub>-C<sub>4</sub> alkyl."

5           The term "halo" refers to chloro, fluoro, bromo or iodo.

          The term "C<sub>2</sub>-C<sub>12</sub> alkenyl" refers to a straight or branched alkenyl chain having from two to twelve carbon atoms. Typical C<sub>2</sub>-C<sub>12</sub> alkenyl groups include ethenyl, 1-  
10   propen-2-yl, 3-buten-1-yl, 1-buten-2-yl, 1-buten-1-yl, 1-penten-3-yl, 2-hexen-3-yl, 1-decen-2-yl, 2-decen-5-yl and the like.

          The term "C<sub>2</sub>-C<sub>12</sub> alkynyl" refers to a straight or branched alkynyl chain having from two to twelve carbon  
15   atoms. Typical C<sub>2</sub>-C<sub>12</sub> alkynyl groups include ethynyl, 1-propyn-1-yl, 1-propyn-2-yl, 1-butyn-1-yl, 1-butyn-3-yl, 1-pentyn-3-yl, 4-pentyn-2-yl, 1-hexyn-3-yl, 3-hexyn-1-yl, 5-methyl-3-hexyn-1-yl, 5-octyn-1-yl, 7-octyn-1-yl, 4-decyn-1-yl, 6-decyn-1-yl and the like.

20           The term "C<sub>1</sub>-C<sub>12</sub> alkylthio" refers to a straight or branched alkyl chain having from one to twelve carbon atoms attached to a sulfur atom. Typical C<sub>1</sub>-C<sub>12</sub> alkylthio groups include methylthio, ethylthio, propylthio, isopropylthio, butylthio, 3-methyl-heptylthio, octylthio,  
25   5,5-dimethyl-hexylthio and the like.

          The term "C<sub>1</sub>-C<sub>12</sub> alkoxy" refers to a straight or branched alkyl chain having from one to twelve carbon atoms attached to an oxygen atom. Typical C<sub>1</sub>-C<sub>12</sub> alkoxy groups include methoxy, ethoxy, propoxy, butoxy, sec-butoxy,  
30   pentoxy, 5-methyl-hexoxy, heptoxy, octyloxy, decyloxy, dodecyloxy and the like. The term "C<sub>1</sub>-C<sub>12</sub> alkyl" includes within its definition the terms "C<sub>1</sub>-C<sub>6</sub> alkoxy" and C<sub>1</sub>-C<sub>4</sub> alkoxy."

          The terms "C<sub>1</sub>-C<sub>12</sub> substituted alkyl," "C<sub>2</sub>-C<sub>12</sub>  
35   substituted alkenyl" and "C<sub>2</sub>-C<sub>12</sub> substituted alkynyl," refers to the specified moiety substituted with 1 or 2 substituents independently selected from halo, hydroxy,



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protected hydroxy, amino, protected amino, C<sub>1</sub>-C<sub>7</sub> acyloxy, nitro, carboxy, protected carboxy, carbamoyl, carbamoyloxy, cyano, methylsulfonylamino, phenyl, substituted phenyl or C<sub>1</sub>-C<sub>12</sub> alkoxy.

5           The term "substituted phenyl" refers to a phenyl group substituted with 1, 2 or 3 substituents independently selected from halo, hydroxy, protected hydroxy, cyano, nitro, C<sub>1</sub>-C<sub>12</sub> alkyl, C<sub>1</sub>-C<sub>12</sub> alkoxy, carboxy, protected  
10 carboxy, carboxymethyl, hydroxymethyl, amino, aminomethyl trifluoromethyl or N-methylsulfonylamino.

          The term "C<sub>3</sub>-C<sub>12</sub> cycloalkyl" refers a saturated hydrocarbon ring structure having from three to twelve carbon atoms. Typical C<sub>3</sub>-C<sub>12</sub> cycloalkyl groups include cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl and  
15 cycloheptyl, cyclooctyl and the like.

          The term "C<sub>3</sub>-C<sub>12</sub> cycloalkoxy" refers to a C<sub>3</sub>-C<sub>12</sub> cycloalkyl group attached to an oxygen atom. Typical C<sub>3</sub>-C<sub>12</sub> cycloalkoxy groups include cyclopropyloxy, cyclobutyloxy, cyclopentyloxy, cyclohexyloxy and  
20 cycloheptyloxy and the like.

          The term "C<sub>3</sub>-C<sub>12</sub> cycloalkenyl" refers to a hydrocarbon ring structure having from three to twelve carbon atoms with at least one double bond. Typical C<sub>3</sub>-C<sub>12</sub> cycloalkenyl groups include cyclopropenyl, cyclobutenyl, cyclopentenyl and the like.  
25

          The term "methyl(C<sub>3</sub>-C<sub>12</sub> cycloalkyl)" refers to a C<sub>3</sub>-C<sub>12</sub> cycloalkyl group that is substituted with a methyl group. Typical methyl(C<sub>3</sub>-C<sub>12</sub> cycloalkyl) groups include 2-methylcyclopropyl, 2-methylcyclobutyl, 3-  
30 methylcyclopentyl, 4-methylcyclohexyl and the like.

          The term "C<sub>1</sub>-C<sub>4</sub> alkylamino" refers to a straight or branched alkylamino chain having from one to four carbon atoms attached to a nitrogen atom. Typical C<sub>1</sub>-C<sub>4</sub> alkylamino groups include methylamino, ethylamino,  
35 propylamino, isopropylamino, butylamino, sec-butylamino and the like.

The term "di(C<sub>1</sub>-C<sub>4</sub> alkyl)amino" refers to a di(C<sub>1</sub>-C<sub>4</sub> alkyl)amino chain having two straight or branched alkyl chains of from one to four carbon atoms attached to a common nitrogen atom. Typical di(C<sub>1</sub>-C<sub>4</sub> alkyl)amino groups  
5 include dimethylamino, diethylamino, ethylmethylamino, methylisopropyl-amino, dipropylamino, dibutylamino, methylbutylamino, *t*-butylisopropylamino, di-*t*-butylamino and the like.

The term "C<sub>2</sub>-C<sub>12</sub> alkanoyl" represents a straight  
10 or branched chain alkyl chain having from one to four carbon atoms attached to a carbonyl moiety. Typical C<sub>2</sub>-C<sub>12</sub> alkanoyl groups include ethanoyl, propanoyl, isopropanoyl, butanoyl, isobutanoyl, sec-butanoyl, *t*-butanoyl, pentanoyl and the like.

The term "C<sub>2</sub>-C<sub>12</sub> alkanoylamino" represents a  
15 straight or branched chain alkyl group attached to a carbonylamino moiety. Typical C<sub>2</sub>-C<sub>12</sub> alkanoylamino groups include ethanoylamino, propanoylamino, isopropanoylamino, butanoyl-amino, isobutanoylamino, sec-butanoylamino, *t*-butanoylamino, pentanoylamino and the like.  
20

The terms "C<sub>7</sub>-C<sub>10</sub> bicycloalkyl" represents two fused cycloalkyl rings having a total of seven to ten carbon atoms and "C<sub>7</sub>-C<sub>14</sub> tricycloalkyl" represents three fused cycloalkyl rings having a total of seven to fourteen  
25 carbon atoms. Typical "C<sub>7</sub>-C<sub>10</sub> bicycloalkyl" and "C<sub>7</sub>-C<sub>14</sub> tricycloalkyl" groups include bicyclo[2.2.1.]hept-2-yl, bicyclo[2.2.1.]hept-4-en-2-yl, bicyclo[3.3.1.]non-3-yl, bicyclo[3.3.1.]non-2-yl, bicyclo[3.2.1.]oct-2-yl, bicyclo[2.2.2.]oct-2-yl, bicyclo[2.2.2.]oct-5-en-2-yl, adamantyl and the like.  
30

The term "hydroxy protecting group" refers to a substituent of an hydroxy group that is commonly employed to block or protect the hydroxy functionality while reactions are carried out on other functional groups on the  
35 compound. Examples of such hydroxy protecting groups include tetrahydropyranyl, 2-methoxyprop-2-yl, 1-ethoxyethyl-yl, methoxymethyl,  $\beta$ -methoxyethoxymethyl,

methythiomethyl, t-butyl, t-amyl, trityl, 4-methoxytrityl, 4,4'-dimethoxytrityl, 4,4',4"-trimethoxytrityl, benzyl, allyl, trimethylsilyl, (t-butyl)dimethylsilyl, and 2,2,2-trichloroethoxycarbonyl and the like. The species of hydroxy protecting group is not critical so long as the derivatized hydroxy group is stable to the conditions of the subsequent reaction(s) and can be removed at the appropriate point without disrupting the remainder of the molecule. A preferred hydroxy protecting group is trimethylsilyl. Further examples of hydroxy protecting groups are described in T.W. Greene, "Protective Groups in Organic Synthesis," John Wiley and Sons, New York, N.Y., (2nd ed., 1991) chapters 2 and 3. The term "protected hydroxy" refers to a hydroxy group bonded to one of the above hydroxy protecting groups.

The term "dideoxy" refers to compounds of the formula I where  $R^{x1}$  and  $R^{x2}$  are each hydrogen.

The term "inhibiting", i.e. a method of inhibiting parasitic or fungal activity, includes stopping, retarding or prophylactically hindering or preventing the growth or any attending characteristics and results from the existence of a parasite or fungus.

The term "contacting", i.e. contacting a compound of the invention with a parasite or fungus, includes a union or junction, or apparent touching or mutual tangency of a compound of the invention with a parasite or fungus. However, the term does not imply any further limitations to the process, such as by mechanism of inhibition, and the methods are defined to encompass the spirit of the invention, which is to inhibit parasitic and fungal activity by the action of the compounds and their inherent antiparasitic and antifungal properties, or in other words, the compounds, used in the claimed methods are the causative agent for such inhibition.

The term "pharmaceutically acceptable salt" as used herein, refers to salts of the compounds of the above formula which are substantially non-toxic to living

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organisms. Typical pharmaceutically acceptable salts include those salts prepared by reaction of the compounds of the present invention with a mineral or organic acid or an inorganic base. Such salts are known as acid addition and base addition salts.

Acids commonly employed to form acid addition salts are mineral acids such as hydrochloric acid, hydrobromic acid, hydroiodic acid, sulfuric acid, phosphoric acid and the like, and organic acids such as *p*-toluenesulfonic, methanesulfonic acid, oxalic acid, *p*-bromophenylsulfonic acid, carbonic acid, succinic acid, citric acid, benzoic acid, acetic acid, and the like. Examples of such pharmaceutically acceptable salts are the sulfate, pyrosulfate, bisulfate, sulfite, bisulfite, phosphate, monohydrogenphosphate, dihydrogenphosphate, metaphosphate, pyrophosphate, chloride, bromide, iodide, acetate, propionate, decanoate, caprylate, acrylate, formate, isobutyrate, caproate, heptanoate, propiolate, oxalate, malonate, succinate, suberate, sebacate, fumarate, maleate, butyne-1,4-dioate, hexyne-1,6-dioate, benzoate, chlorobenzoate, methylbenzoate, dinitrobenzoate, hydroxybenzoate, methoxybenzoate, phthalate, sulfonate, xylenesulfonate, phenylacetate, phenylpropionate, phenylbutyrate, citrate, lactate,  $\gamma$ -hydroxybutyrate, glycollate, tartrate, methanesulfonate, propanesulfonate, naphthalene-1-sulfonate, naphthalene-2-sulfonate, mandelate and the like. Preferred pharmaceutically acceptable acid addition salts are those formed with mineral acids such as hydrochloric acid and hydrobromic acid, and those formed with organic acids such as maleic acid and methanesulfonic acid.

Base addition salts include those derived from inorganic bases, such as ammonium or alkali or alkaline earth metal hydroxides, carbonates, bicarbonates, and the like. Such bases useful in preparing the salts of this invention thus include sodium hydroxide, potassium hydroxide, ammonium hydroxide, potassium carbonate, sodium

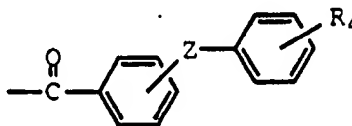
-11-

carbonate, sodium bicarbonate, potassium bicarbonate, calcium hydroxide, calcium carbonate, and the like. The potassium and sodium salt forms are particularly preferred.

It should be recognized that the particular counterion forming a part of any salt of this invention is not of a critical nature, so long as the salt as a whole is pharmacologically acceptable and as long as the counterion does not contribute undesired qualities to the salt as a whole.

Typical examples of acyl groups at  $R_2$  in formula I include benzoyl substituted by a polyoxa-alkyl group such as 2-methoxyethoxy (p is 0, m is 1), 2-ethoxyethoxy (p is 0, m is 2), 2-(2-ethoxyethoxy)ethoxy (m is 2, p is 1, n is 2), 3-(2-ethoxyethoxy)propoxy, 4-(2-methoxyethoxy)butoxy, and the like, or benzoyl substituted by alkynyl groups ( $-C\equiv C-(C_1-C_{12} \text{ alkyl})$ ) such as propynyl, butynyl, hexynyl, undecynyl, and the like, or *cis* or *trans* alkenyl groups ( $-CH_2=CH_2-(C_1-C_{12} \text{ alkyl})$ ) such as propenyl, butenyl, hexenyl, decenyl, and the like.

Examples of acyl groups where  $R_2$  is a group of the formula



include diphenyl ethers (Z is  $-O-$ ), diphenyl acetylenes (Z is  $-C\equiv C-$ ), stilbenes (Z is  $-CH=CH-$ ) and biphenyls (Z is a bond).

Examples of diphenyl ether groups include 4-(4-butoxyphenoxy)benzoyl, 4-(4-hexoxyphenoxy)benzoyl, 4-(4-ethoxyphenoxy)benzoyl, 4-(4-phenyloxyphenoxy)benzoyl, 4-[4-(4-dodecyloxyphenoxy)benzoyl, 4-[4-(3-dimethylaminopropoxy)phenoxy]benzoyl and the like.

Examples of diphenyl acetylene and stilbene groups include 4-styrylbenzoyl, 4-(4-methoxystyryl)benzoyl, 4-(4-butoxystyryl)benzoyl, 4-(phenylethynyl)benzoyl, 4-(4-ethoxyphenylethynyl)benzoyl, 4-(4-cyclohexyloxyphenylethynyl)benzoyl and the like.

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Examples of biphenyl groups include 4-[4-(butoxy)phenyl]benzoyl, 4-[4-(cyclobutylmethoxy)phenyl]benzoyl, 4-[4-(cyclopentylmethoxy)phenyl]benzoyl, 4-[4-(cyclohexylethoxy)phenyl]benzoyl, 4-[4-(hexoxy)phenyl]benzoyl, 4-phenylbenzoyl, 4-[4-(11-amino-undecyloxy)phenyl]benzoyl, 4-[4-(11-formamidoundecyloxy)phenyl]benzoyl, 4-[4-(iso-pentoxy)phenyl]benzoyl and the like.

Examples of biphenyl groups where  $R_4$  is  $-O-(CH_2)_r-W-R_5$  include 4-[4-[2-(N-cyclohexylpiperidino-4-yl)ethoxy]phenyl]benzoyl, 4-[4-[2-(N-hexylpiperidino-4-yl)ethoxy]phenyl]benzoyl, 4-[4-[2-(4-benzylpiperidino)ethoxy]phenyl]benzoyl, 4-[4-[2-(4-cyclohexylpiperidino)ethoxy]phenyl]benzoyl and the like.

Examples of biphenyl and diphenyl ether groups where  $R_4$  is  $-Y^1-R_6$  include 4-[4-(phenylethynyl)phenyl]benzoyl, 4-[4-(phenylethynyl)phenoxy]benzoyl, 4-[4-(hexynyl)phenyl]benzoyl, 4-[4-(styryl)phenoxy]benzoyl, 4-[4-[4-(4-methylpiperidino)ethoxy]phenylethynyl]phenyl]benzoyl, and the like.

Acyl groups where  $R_4$  is  $-O-(CH_2)_r-W-R_5$  may form acid addition salts of the basic amino groups of the piperidine and piperazine heterocyclic groups with organic or mineral acids such as hydrochloric acid, hydrobromic acid, sulfuric acid and phosphoric acid and with organic acids such as the sulfonic acids, benzenesulfonic acid, toluenesulfonic acid, methanesulfonic acid, acetic acid, chloroacetic acid, trifluoroacetic acid, benzoic acid, isophthalic acid, salicylic acid, citric acid, malic acid, succinic acid, malonic acid and the like.

Table 1, below, provides further examples of acyl groups,  $R_2$ , found on cyclic peptides of formula I.

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Table 1

<u>R<sub>2</sub></u>

5

Table 1 (continued)

<u>R<sub>2</sub></u>

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-14-

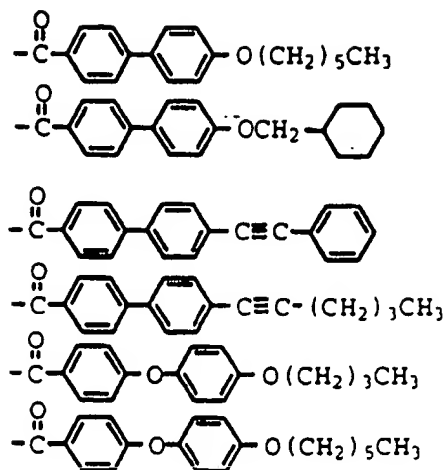
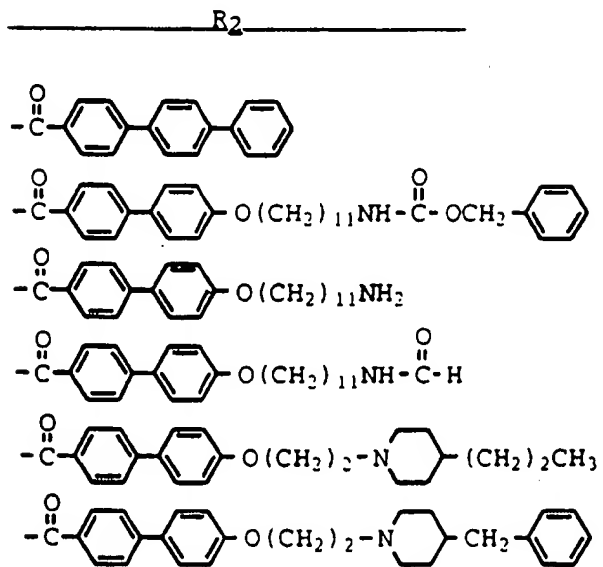


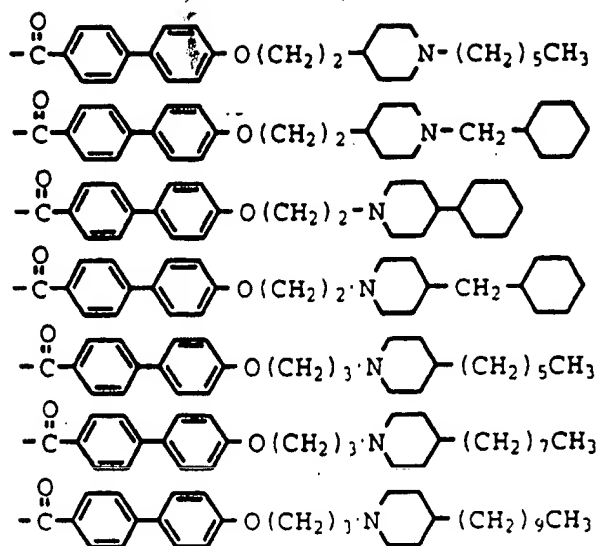
Table 1 (continued)

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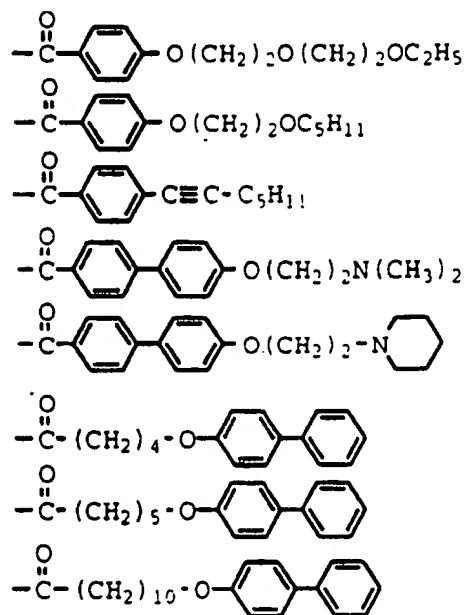




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Table 1 (continued)

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R<sub>2</sub>

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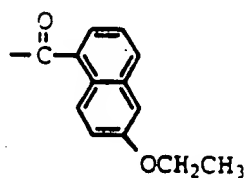
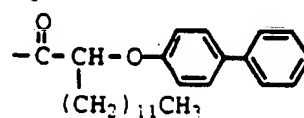
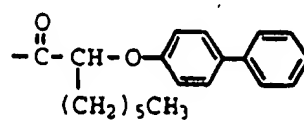
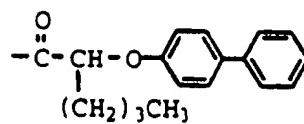
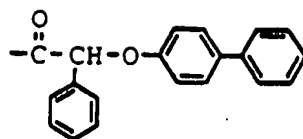
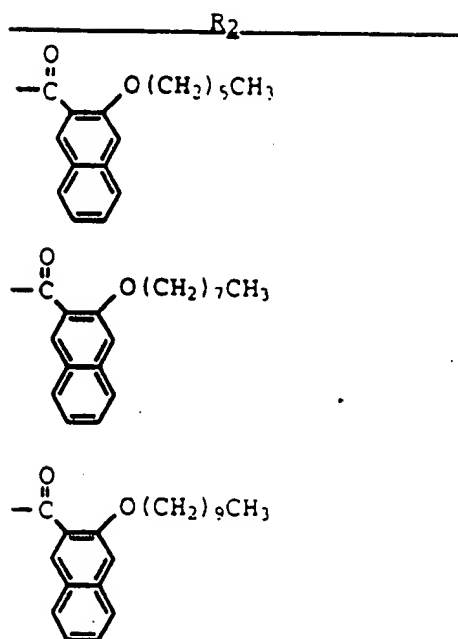
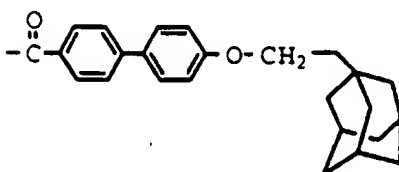
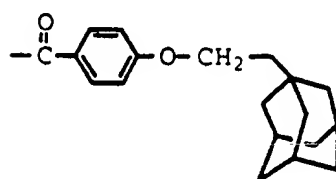
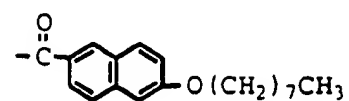
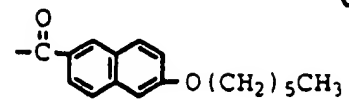
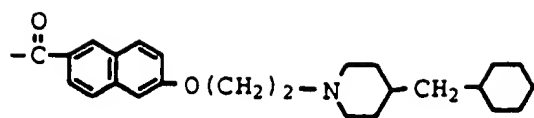


Table 1 (continued)

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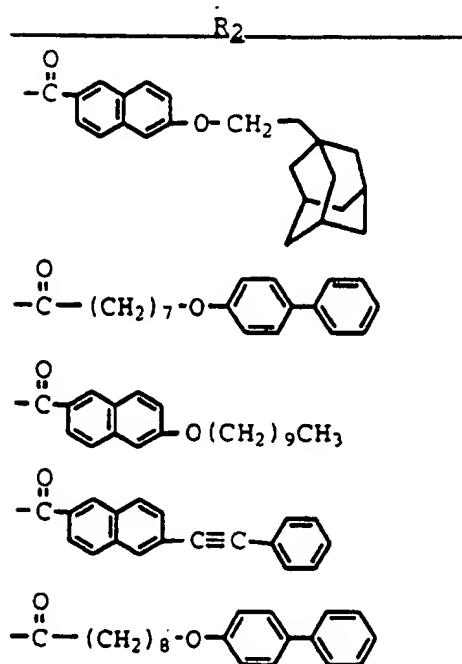


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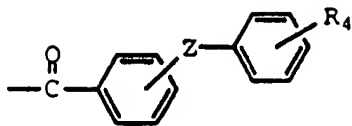
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Table 1 (continued)



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Preferred acyl groups,  $R_2$ , include groups of the formula:



5

wherein:

Z is  $\text{---C}\equiv\text{C---}$ ,  $\text{---CH=CH---}$ ,  $\text{---CH}_2\text{---CH}_2\text{---}$ , or a bond;

A)  $R_4$  is hydrogen,  $\text{C}_1\text{---C}_{12}$  alkyl,  $\text{C}_1\text{---C}_{12}$  substituted alkyl,  $\text{C}_2\text{---C}_{12}$  alkenyl,  $\text{C}_2\text{---C}_{12}$  substituted alkenyl,  $\text{C}_2\text{---C}_{12}$  alkynyl,  $\text{C}_2\text{---C}_{12}$  substituted alkynyl,  $\text{C}_1\text{---C}_{12}$  alkoxy,  $\text{C}_3\text{---C}_{12}$  cycloalkyl,  $\text{C}_7\text{---C}_{10}$  bicycloalkyl,  $\text{C}_7\text{---C}_{14}$  tricycloalkyl,  $\text{C}_3\text{---C}_{12}$  cycloalkoxy, naphthyl, pyridyl, thienyl, benzothienyl, quinolyl or phenyl; or

B)  $R_4$  is phenyl substituted by amino,  $\text{C}_1\text{---C}_{12}$  alkylthio, halo,  $\text{C}_1\text{---C}_{12}$  alkyl,  $\text{C}_2\text{---C}_{12}$  alkenyl,  $\text{C}_2\text{---C}_{12}$  alkynyl,  $\text{C}_1\text{---C}_{12}$  substituted alkyl,  $\text{C}_2\text{---C}_{12}$  substituted alkenyl,  $\text{C}_2\text{---C}_{12}$  substituted alkynyl,  $\text{C}_1\text{---C}_{12}$  alkoxy, trifluoromethyl, phenyl, substituted phenyl, or a group of the formula  $\text{---O---(CH}_2\text{)}_m\text{---[O---(CH}_2\text{)}_n\text{]}_p\text{---O---(C}_1\text{---C}_{12}\text{ alkyl)}$  where m, n and p are as defined above; or

C)  $R_4$  is  $\text{C}_1\text{---C}_{12}$  alkoxy substituted with halo,  $\text{C}_3\text{---C}_{12}$  cycloalkyl,  $\text{C}_7\text{---C}_{10}$  bicycloalkyl,  $\text{C}_7\text{---C}_{14}$  tricycloalkyl,  $\text{C}_1\text{---C}_6$  alkoxy,  $\text{C}_2\text{---C}_{12}$  alkynyl, amino,  $\text{C}_1\text{---C}_4$  alkylamino, di( $\text{C}_1\text{---C}_4$  alkyl)amino, formamido,  $\text{C}_2\text{---C}_{12}$  alkanoylamino, or phenyl substituted with a group of the formula  $\text{---O---(CH}_2\text{)}_m\text{---[O---(CH}_2\text{)}_n\text{]}_p\text{---O---(C}_1\text{---C}_{12}\text{ alkyl)}$  where m, n and p are as defined above; or

30

D)  $R_4$  is  $\text{---O---(CH}_2\text{)}_r\text{---W---R}_5$ ;

r is 2, 3 or 4;

W is pyrrolidino, piperidino or piperazino;

$R_5$  is hydrogen,  $\text{C}_1\text{---C}_{12}$  alkyl,  $\text{C}_3\text{---C}_{12}$  cycloalkyl, benzyl or  $\text{C}_3\text{---C}_{12}$  cycloalkylmethyl; or

35

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E)  $R_4$  is  $-Y^1-R_6$ ;

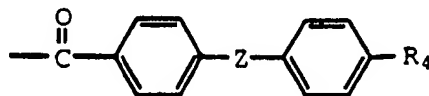
$Y^1$  is  $-C\equiv C-$  or  $-CH=CH-$ ;

$R_6$  is  $C_3-C_{12}$  cycloalkyl,  $C_7-C_{10}$  bicycloalkyl,  
 5  $C_7-C_{14}$  tricycloalkyl,  $C_3-C_{12}$  cycloalkenyl, naphthyl,  
 benzothiazolyl, thienyl, indanyl, fluorenyl, or phenyl  
 substituted with  $C_1-C_{12}$  alkylthio,  $C_2-C_{12}$  alkenyl,  $C_2-C_{12}$   
 alkynyl, halo( $C_1-C_6$  alkoxy) or a group of the formula  
 $-O-(CH_2)_r-W-R_5$  where  $r$ ,  $W$  and  $R_5$  are as defined above; or  
 10  $R_6$  is phenyl substituted with a group of the  
 formula  $-O-(CH_2)_m-[O-(CH_2)_n]_p-O-(C_1-C_{12} \text{ alkyl})$  where  $m$ ,  $n$   
 and  $p$  are as defined above; or

F)  $R_4$  is  $C_1-C_{12}$  alkoxy substituted with a group of  
 15 the formula  $-NHC(O)R_7$ ;

$R_7$  is  $C_1-C_6$  alkoxy, or phenyl( $C_1-C_6$  alkoxy);  
 or a pharmaceutically acceptable salt thereof.

More preferred are acyl groups,  $R_2$ , of the formula:  
 20

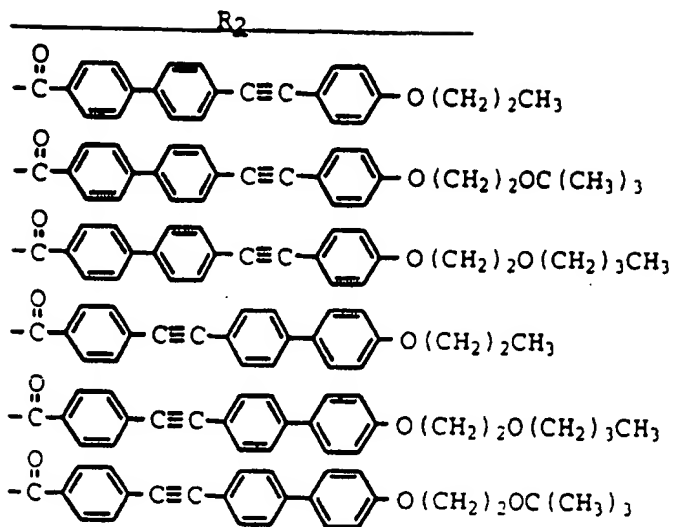


where  $Z$  is  $-C\equiv C-$  or a bond;  
 or a pharmaceutically acceptable salt thereof.

25 Table 2, below, provides a list of preferred  
 acyl groups,  $R_2$ , found on cyclic peptides of formula I.

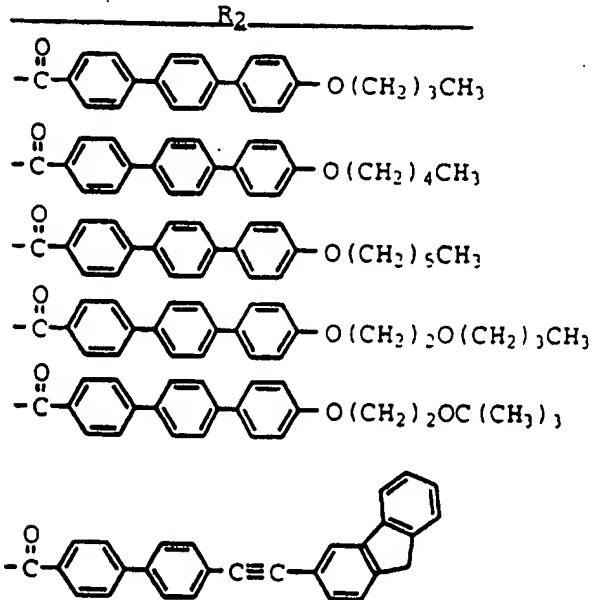
-20-

Table 2



5

Table 2 (continued)



10

Preferred compounds of this invention are those compounds of formula I where:

R', R'', and R''' are each methyl;

RY<sup>1</sup>, RY<sup>2</sup>, RY<sup>3</sup>, RY<sup>4</sup> are each hydroxy;

15

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or a pharmaceutically acceptable salt thereof.

Of these compounds, more preferred are those compounds of formula I where:

5

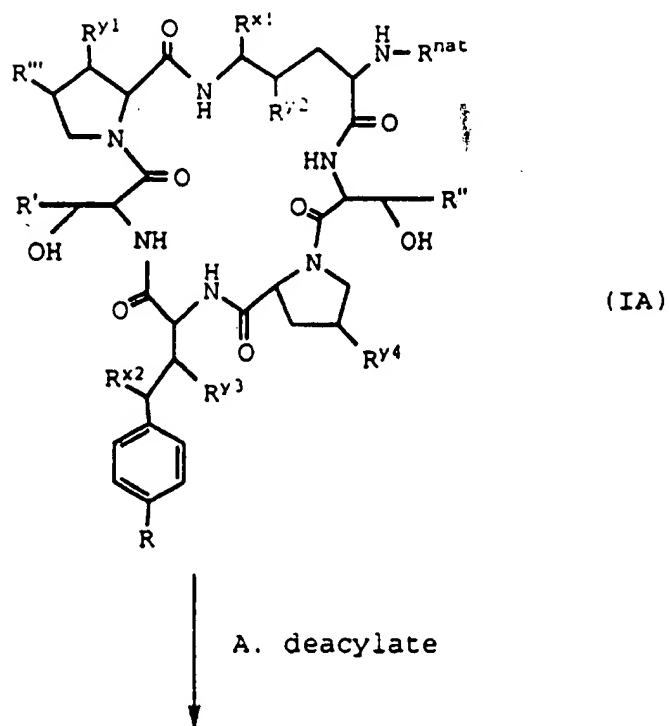
$R^{x1}$  and  $R^{x2}$  are hydrogen;

or a pharmaceutically acceptable salt thereof.

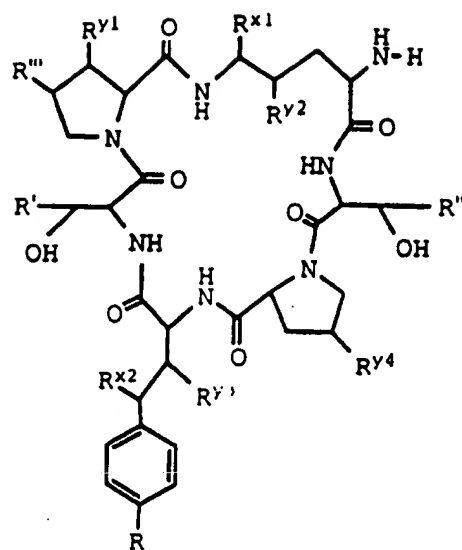
The compounds of formula I may be prepared according to Reaction Scheme I, as follows.

10

Reaction Scheme I

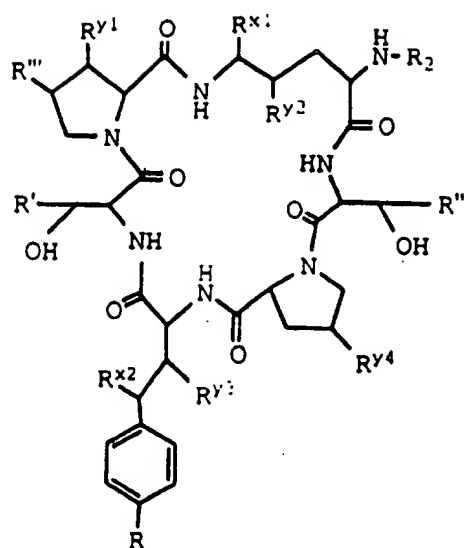


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(IB)

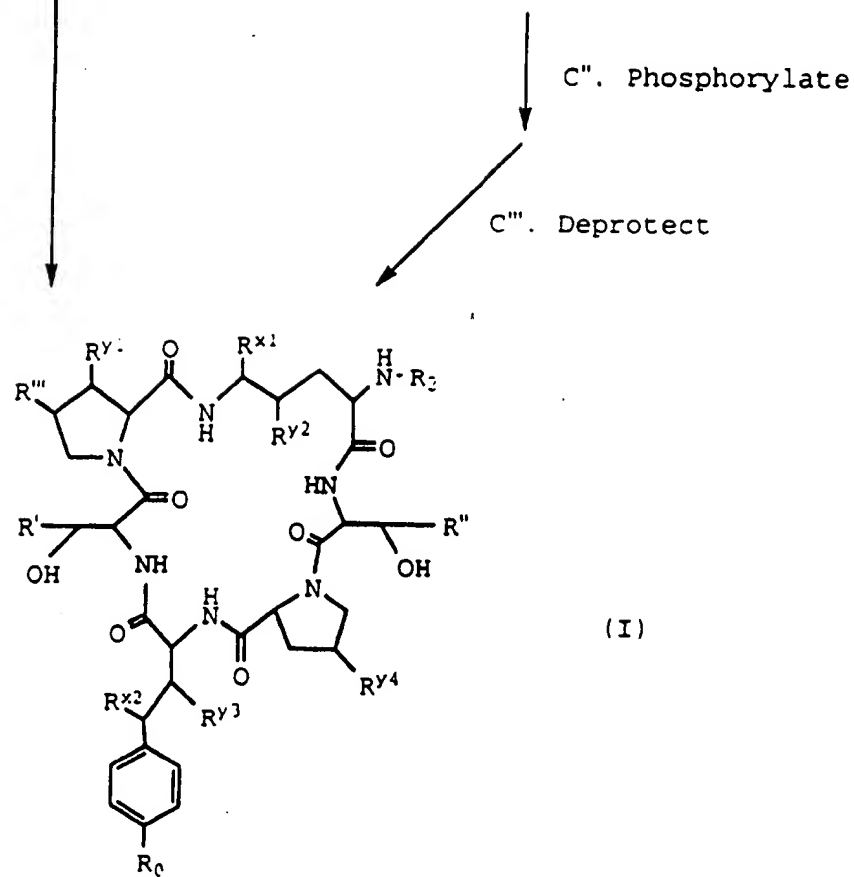
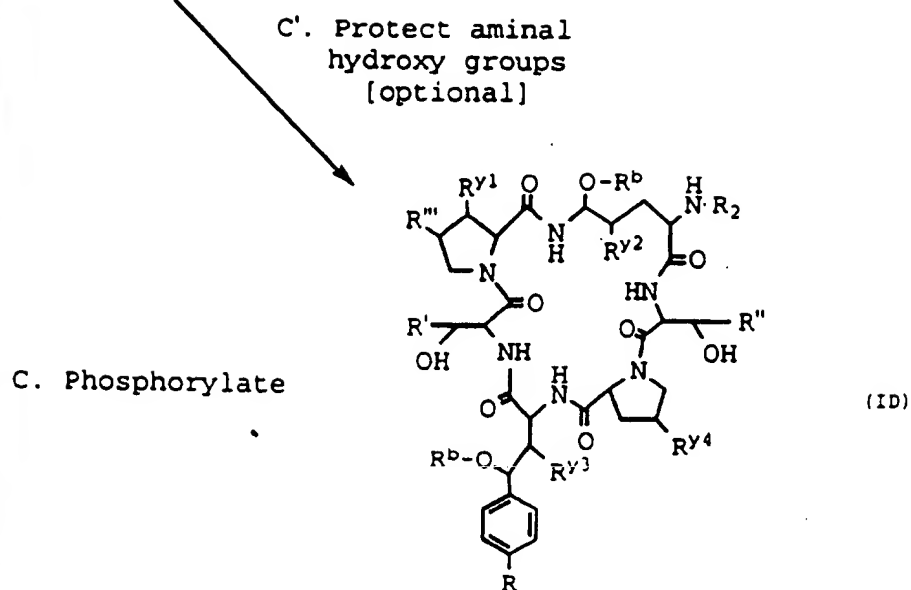
B. re-acylate



(IC)



-23-



wherein:

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R<sup>nat</sup> is a naturally occurring cyclic peptide sidechain;

R is hydroxy;

R<sup>b</sup> is an hydroxy protecting group; and

5 R', R'', R''', R<sup>x1</sup>, R<sup>x2</sup>, RY<sup>1</sup>, RY<sup>2</sup>, RY<sup>3</sup>, RY<sup>4</sup>, R, R<sub>0</sub> and R<sub>2</sub> are as defined above.

Reaction scheme I, above, is accomplished by carrying out reactions A-C (or A-C'''), in order. Once a reaction is complete, the intermediate compound may be  
10 isolated by procedures well-known in the art, for example, the compound may be crystallized and then collected by filtration, or the reaction solvent may be removed by extraction, evaporation or decantation. The intermediate compound may be further purified, if desired, by common  
15 techniques such as crystallization or chromatography over solid supports such as silica gel, alumina and the like, before carrying out the next step of the reaction scheme.

In reaction IA, a naturally occurring cyclic peptide of the formula (IA) is deacylated using procedures  
20 known in the art to provide an amino nucleus of formula (IB). This reaction is typically carried out using enzymatic acylation by exposing the naturally occurring cyclic peptide to a deacylase enzyme. The deacylase enzyme may be obtained from the microorganism Actinoplanes  
25 utahensis and used substantially as described in U.S. Patent Nos. 4,293,482 and 4,304,716, herein incorporated by reference. The deacylase enzyme may also be obtained from the Pseudomonas species. Deacylation may be accomplished using whole cells of Actinoplanes utahensis  
30 or Pseudomonas or the crude or purified enzyme thereof or using an immobilized form of the enzyme. See European Patent Application No. 0 460 882 (December 11, 1991). Examples of naturally occurring cyclic peptides which may be used as starting materials include aculeacin (palmitoyl  
35 side chain), tetrahydroechinocandin B (stearoyl side chain), mulundocandin (branched C<sub>15</sub> side chain), L-671,329 (C<sub>16</sub> branched side chain), S 31794/F1 (tetradecanoyl side

-25-

chain), sporiofungin ( $C_{15}$  branched side chain), FR901379 (palmitoyl side chain) and the like. A preferred naturally occurring cyclic peptide is echinocandin B (a compound of formula (IA) where  $R'$ ,  $R''$ , and  $R'''$  are each methyl,  $R^{x1}$  and  $R^{x2}$  are each hydroxy,  $RY^1$ ,  $RY^2$ ,  $RY^3$ ,  $RY^4$  are each hydroxy,  $R$  is hydroxy and  $R_2$  is linoleoyl).

In Reaction IB, the resulting amino nucleus is then re-acylated using procedures known in the art to provide a compound of formula I where  $R_2$  is an acyl group as defined hereinabove.

For example, the amino nucleus may be acylated by reaction with an appropriately substituted acyl halide, preferably in the presence of an acid scavenger such as a tertiary amine, such as triethylamine. The reaction is typically carried out at a temperature of from about  $-20^\circ\text{C}$  to about  $25^\circ\text{C}$ . Typical solvents for this reaction include polar aprotic solvents such as dioxane or dimethylformamide. Solvent choice is not critical so long as the solvent employed is inert to the ongoing reaction and the reactants are sufficiently solubilized to effect the desired reaction.

The amino nucleus may also be acylated by reaction with an appropriately substituted carboxylic acid, in the presence of a coupling agent. Typical coupling agents include dicyclohexylcarbodiimide (DCC),  $N,N'$ -carbonyldiimidazole, bis(2-oxo-3-oxazolidinyl)phosphinic chloride (BOP-Cl),  $N$ -ethoxycarbonyl-2-ethoxy-1,2-dihydroquinoline (EEDQ), benzotriazol-1-yloxytripyrrolidinophosphonium hexafluorophosphate (PyBOP) and the like.

In addition, the amino nucleus may be acylated with an activated ester of a carboxylic acid ( $\text{RCOOH}$ ) such as an ester of a carboxylic acid of the formula  $R_2\text{-COOH}$  and  $p$ -nitrophenyl, 2,4,5-trichlorophenyl, hydroxybenzotriazole hydrate ( $\text{HOBT}\cdot\text{H}_2\text{O}$ ), pentafluorophenol,  $N$ -hydroxysuccinimide and the like. Preferred acylating moieties are the active esters of the carboxylic acid  $R_2\text{-COOH}$  such as 2,4,5-

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trichlorophenyl ester and HOBT ester. The reaction is typically carried out for one to sixty five hours at a temperature from about 0°C to about 30°C in an aprotic solvent. The reaction is generally complete after about  
5 twenty four to forty eight hours when carried out at a temperature of from about 15°C to about 30°C. Typical solvents for this reaction are tetrahydrofuran and dimethylformamide or a mixture of such solvents. The amino nucleus is generally employed in equimolar proportions  
10 relative to the activated ester or with a slight excess of the amino nucleus.

In Reaction IC, the compound of formula (IC) is phosphorylated by reaction with an appropriately substituted alkyl or phenyl phosphate to provide a compound  
15 of formula I where  $R_0$  is  $P(O)_2OH-R_1$  where  $R_1$  is  $C_1-C_6$  alkoxy or phenoxy, or by reaction with an appropriately substituted alkyl or phenyl phosphonic acid to provide a compound of formula I where  $R_0$  is  $-P(O)_2OH-R_1$  where  $R_1$  is  $C_1-C_6$  alkyl, an appropriately substituted phenyl or benzyl  
20 moiety. The reaction is carried out in the presence of a base such as lithium trimethyl-silanolate (LiOTMS), lithium bis(trimethylsilyl)amide (LHMDS), pyridine and the like. The reaction is typically carried out for up to one hour at a temperature from about -30°C to about 0°C in an aprotic  
25 solvent such as tetrahydrofuran and dimethylformamide. The reaction is generally complete in about fifteen minutes when carried out under these conditions. The phosphate or phosphonate reactant is generally employed in equimolar proportions to about a one mole excess relative to the amino nucleus in the presence of an equimolar or slight  
30 excess of the base. Phosphorylation of an amino nucleus with unprotected amination hydroxy groups ( $R^{x1}$  and  $R^{x2}$ ) is typically carried out at a temperature from about -30°C to about -15°C.

35 Alternatively, in Reaction IC' the acylated nucleus of formula (IC), where  $R^{x1}$  and/or  $R^{x2}$  are hydroxy, may be optionally protected with an hydroxy protecting

group using procedures known in the art. For example, the reaction is typically carried out by combining the compound of formula (IC) with a suitable hydroxy protecting group in the presence of a catalyst at a temperature in the range of from about 0°C to about 40°C for about one to five hours in a mutual inert solvent. The hydroxy protecting group is generally employed in an amount ranging from about equimolar proportions to about a 100 molar excess relative to the compound of formula (IC), preferably in a large molar excess. Suitable catalysts include strong acids such as p-toluenesulfonic acid, camphorsulfonic acid (CSA), hydrochloric acid, sulfuric acid, trifluoroacetic acid and the like. Typical solvents suitable for use in this reaction include any organic solvent such as dioxane. Solvent choice is not critical so long as the solvent employed is inert to the ongoing reaction and the reactants are sufficiently solubilized to effect the desired reaction. The reaction is preferably conducted at a temperature in the range of from about 20°C to about 30°C for about 2 to 4 hours. It is not necessary to protect the dideoxy compounds of formula (IC), that is, those compounds of formula (IC) where  $R^{x1}$  and  $R^{x2}$  are hydrogen.

Reaction IC" is carried out as described above in Reaction IC, above.

In Reaction IC"', the hydroxy protecting groups that were used to protect the aminor hydroxy moieties in Reaction IC', above, are removed according to procedures known in the art to provide the desired compound of formula I. For example, the protecting groups can be removed by reaction with a Lewis acid in a mutual inert organic solvent such as methylene chloride. Examples of Lewis acids include trimethylsilylbromide, boron trifluoride etherate and the like. The reaction is typically carried out at a temperature of from about 0°C to about 40°C, preferably at a temperature of from about 20°C to about 30°C. A preferred Lewis acid is boron trifluoride etherate.

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The cyclic peptides used to make the compounds of the present invention may be prepared by fermentation of known microorganisms. For example, the cyclic peptide of formula (IB) where R', R'', and R''' are methyl, R<sup>x1</sup> and R<sup>x2</sup> are hydroxy, RY<sup>1</sup>, RY<sup>2</sup>, RY<sup>3</sup>, RY<sup>4</sup> are hydroxy, and R is hydroxy (cyclic nucleus corresponding to A-30912A) may be prepared using the procedure detailed in Abbott et al., U.S. Pat. Ser. No. 4,293,482, which is herein incorporated by reference. The cyclic peptide of formula (IB) where R', R'', and R''' are methyl, R<sup>x1</sup> is hydroxy, R<sup>x2</sup> is hydrogen, RY<sup>1</sup>, RY<sup>2</sup>, RY<sup>3</sup>, RY<sup>4</sup> are hydroxy and R is hydroxy (cyclic nucleus corresponding to A-30912B) may be prepared using the procedure detailed in Abbott et al., U.S. Pat. Ser. No. 4,299,763, which is herein incorporated by reference. The cyclic peptide of formula (IB) where R', R'', and R''' are methyl, R<sup>x1</sup> and R<sup>x2</sup> are hydrogen, RY<sup>1</sup>, RY<sup>3</sup> and RY<sup>4</sup> are hydroxy, RY<sup>2</sup> is hydrogen and R is hydroxy (cyclic nucleus corresponding to A-30912D) may be prepared using the procedure detailed in Abbott et al., U.S. Pat. Ser. No. 4,299,762, which is herein incorporated by reference. Aculeacin may be prepared using the procedure detailed in Mizuno et al., U.S. Pat. Ser. No. 3,978,210 which is herein incorporated by reference. The cyclic peptide of formula (IB) where R' is -CH<sub>2</sub>C(O)NH<sub>2</sub>, R'' is methyl, R''' is hydrogen, R<sup>x1</sup>, R<sup>x2</sup>, RY<sup>1</sup>, RY<sup>3</sup>, RY<sup>4</sup> are hydroxy, R is hydroxy may be prepared by deacylating the cyclic peptide prepared using the procedure detailed in Shieh-Shung et al., U.S. Pat. Ser. No. 5,198,421, which is herein incorporated by reference.

The dideoxy compounds of formula I are prepared by removing the benzylic and aminor hydroxy groups (R<sup>x2</sup> and R<sup>x1</sup>, respectively). The hydroxy groups may be removed by subjecting a non-dideoxy compound of formula I (where R<sub>2</sub> is hydrogen or acyl) to a strong acid and a reducing agent at a temperature of between -5°C and 70°C, in a suitable solvent. Typical strong acids include trichloroacetic acid, trifluoroacetic acid or borontrifluoride etherate. A

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preferred strong acid is trifluoroacetic acid. Typical reducing agents include sodium cyanoborohydride or triethylsilane. A preferred reducing agent is triethylsilane. Suitable solvents include methylene chloride, chloroform or acetic acid, preferably methylene chloride. The strong acid should be present in an amount of from 2 to 60 mol per mol of substrate, and the reducing agent should be present in an amount of 2 to 60 mol per mol of substrate. This process affords selective removal of the amination and benzylic hydroxy groups.

The  $R_2$ -COOH precursor acids are prepared by hydrolyzing a nitrile of the formula  $R_2$ -CN or an ester of the formula  $R_2$ -COO( $C_1$ - $C_4$  alkyl). The nitrile and ester intermediates may be prepared using procedures known in the art.

For example, the nitrile and ester intermediates where  $R_2$  is an alkoxy aryl moiety may be prepared using Procedure A or Procedure B, described below.

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#### Procedure A

One equivalent of an alkyl bromide, iodide, or p-toluenesulfonate is added to a mixture containing one equivalent of a base, such as potassium t-butoxide or potassium carbonate, and one equivalent of an hydroxy aryl compound in 200-300 ml of acetonitrile. The resulting reaction mixture is refluxed for approximately six hours and then concentrated in vacuo to provide a residue. This residue is dissolved in a mixture of diethyl ether and a 2N sodium hydroxide solution. The resulting layers are separated and the organic layer is dried over magnesium sulfate, filtered and dried to provide the desired alkoxy aryl product.

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#### Procedure B

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One equivalent of diethylazodicarboxylate is added dropwise over ten minutes, at room temperature, to a mixture containing one equivalent of an hydroxy aryl

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compound, one equivalent of an alkyl alcohol and one equivalent of triphenylphosphine in 200-300 ml of tetrahydrofuran. After approximately seventeen hours, the solvent is removed in vacuo to provide a residue. This residue is dissolved in diethyl ether and the resulting mixture is washed with a 2N sodium hydroxide solution, dried over magnesium sulfate, filtered and concentrated to provide a product which is then crystallized from a diethyl ether/pentane mixture or, if the product contains a tertiary amine, the hydrochloride salt is formed and crystallized from a methanol/ethyl acetate mixture.

The nitrile and ester intermediates where  $R_2$  is an alkynyl or alkenyl aryl moiety may be prepared using Procedure C, below.

#### Procedure C

A mixture containing two equivalents of triethylamine, 0.05 equivalent of palladium dichloride, 0.1 equivalent of triphenylphosphine, 0.025 equivalent of cuprous iodide and one equivalent of an alkyne or two equivalents of an alkene, is added to one equivalent of an aryl bromide, iodide, or trifluoromethanesulfonate in acetonitrile (600 ml/0.1 mol of aryl reactant), under nitrogen. The resulting mixture is refluxed for approximately seventeen hours and then the solvent is removed in vacuo to provide a residue. This residue is slurried in 300 ml of diethyl ether and then filtered to remove the resultant solids. The filtrate is washed with a 1N hydrochloric acid solution, dried over magnesium sulfate, filtered and then dried to provide the desired product.

The ester intermediates where  $R_2$  is a terphenyl moiety may be prepared using Procedure D, below.



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Procedure D1. Formation of boronic acid reactant

Butyl lithium (1.2 equivalents) is added to one equivalent of a cold (-78°C) aryl halide in tetrahydrofuran. After approximately fifteen minutes, two equivalents of triisopropyl borate are added. After approximately ten minutes, the reaction mixture is warmed to room temperature, and then quenched by the addition of water, followed by the addition of a 1N hydrochloric acid solution. The resulting layers are separated and the organic layer is concentrated *in vacuo* to provide a solid. This solid is collected by filtration and then washed with hexane to provide a pure boronic acid product.

2. Formation of terphenyl ester

Tetrakis(triphenylphosphine)palladium (0.03 equivalent) is added to a mixture containing one equivalent of an aryl boronic acid, 1.5 equivalents of potassium carbonate and one equivalent of methyl 4-iodobenzoate (or trichlorophenyl ester of iodobenzoate) in nitrogen-purged toluene. The resulting reaction mixture is refluxed for approximately seven hours and then decanted to remove the potassium carbonate and dried *in vacuo* to provide a residue. This residue is triturated in acetonitrile and then filtered to provide the desired solid product.

The aryl nitriles and esters described above may be converted to the corresponding carboxylic acids by hydrolysis using Procedure E or Procedure F, below.

Procedure E

An aryl nitrile is dissolved in ethanol and an excess of 50% sodium hydroxide solution and refluxed for approximately two hours. Water is added to the resulting reaction mixture until a solid precipitates. This solid is collected by filtration, added to a dioxane/6N hydrochloric

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acid mixtur and the resulting mixture is refluxed for approximately seventeen hours. When the reaction is substantially complete, the carboxylic acid product is crystallized by the addition of water and then collected by filtration and dried *in vacuo*.

#### Procedure F

An excess of a 2N sodium hydroxide solution is added to an aryl ester in methanol, and the resulting solution is refluxed for approximately five hours and then acidified by the addition of excess hydrochloric acid. Water is added to the resulting reaction mixture until a solid (carboxylic acid) precipitates. The carboxylic acid is collected by filtration and dried *in vacuo*.

The carboxylic acids may be converted to the corresponding 2,4,5-trichlorophenyl esters using Procedure G, below. These activated esters are then used to acylate the amino nucleus, as described above in Reaction Scheme IC.

#### Procedure G

A mixture containing one equivalent of an aryl carboxylic acid, one equivalent of 2,4,5-trichlorophenol, and one equivalent of N,N'-dicyclohexylcarbodiimide (DCC) in methylene chloride is stirred for approximately seventeen hours and then filtered. The filtrate is concentrated to provide a residue. This residue is dissolved in diethyl ether, filtered, and pentane is added until crystallization begins. The crystalline product is collected by filtration and dried *in vacuo*.

The following Preparations and Examples further describe how to synthesize the compounds of the present invention. The terms melting point, proton nuclear magnetic resonance spectra, mass spectra, infrared spectra, ultraviolet spectra, elemental analysis, high performance

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liquid chromatography, and thin layer chromatography are abbreviated "m.p.", "NMR", "MS", "IR", "UV", "Analysis", "HPLC", and "TLC", respectively. In addition, the absorption maxima listed for the IR spectra are only those of interest and not all of the maxima observed.

In conjunction with the NMR spectra, the following abbreviations are used: "s" is singlet, "d" is doublet, "dd" is doublet of doublets, "t" is triplet, "q" is quartet, "m" is multiplet, "dm" is a doublet of multiplets and "br.s", "br.d", "br.t", and "br.m" are broad singlet, doublet, triplet, and multiplet respectively. "J" indicates the coupling constant in Hertz (Hz). Unless otherwise noted, NMR data refers to the free base of the subject compound.

The nuclear magnetic resonance spectra were obtained on a General Electric QE-300 300 MHz instrument. The chemical shifts are expressed in delta ( $\delta$ ) values (parts per million downfield from tetramethylsilane).

#### Preparation 1

The following nitrile and ester intermediates where R<sub>2</sub> is an alkoxy aryl moiety were prepared substantially in accordance with Procedure A, detailed above.

Table A


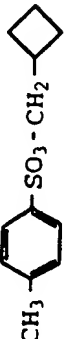





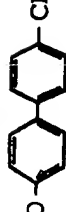
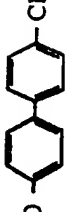
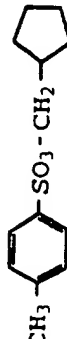


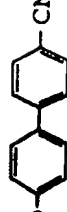
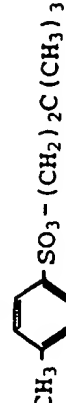





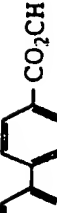










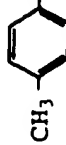



Alkyl halide or tosylate	mass (g)	Alkoxy aryl product	mass (g)
I-(CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	9.4	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> O- 	3.2
CH <sub>3</sub> -  -SO <sub>3</sub> -CH <sub>2</sub> - 	12.3	 -CH <sub>2</sub> O- 	5.3
Br-(CH <sub>2</sub> ) <sub>2</sub> CH(CH <sub>3</sub> ) <sub>2</sub>	7.7	(CH <sub>3</sub> ) <sub>2</sub> CH(CH <sub>2</sub> ) <sub>2</sub> O- 	9.2
CH <sub>3</sub> -  -SO <sub>3</sub> -(CH <sub>2</sub> ) <sub>2</sub> O(CH <sub>2</sub> ) <sub>4</sub> CH <sub>3</sub>	7.6	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>4</sub> O-(CH <sub>2</sub> ) <sub>2</sub> O- 	4.8
Br-(CH <sub>2</sub> ) <sub>4</sub> CH <sub>3</sub>	15.3	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>4</sub> O- 	20.3
CH <sub>3</sub> -  -SO <sub>3</sub> -CH <sub>2</sub> - 	13.0	 -CH <sub>2</sub> O- 	12.2
CH <sub>3</sub> -  -SO <sub>3</sub> -(CH <sub>2</sub> ) <sub>2</sub> C(CH <sub>3</sub> ) <sub>3</sub>	13.1	(CH <sub>3</sub> ) <sub>3</sub> C(CH <sub>2</sub> ) <sub>2</sub> O- 	11.8
Br-CH <sub>2</sub> CH(CH <sub>2</sub> CH <sub>3</sub> ) <sub>2</sub>	8.5	(CH <sub>2</sub> CH <sub>3</sub> ) <sub>2</sub> CHCH <sub>2</sub> O- 	3.0
I-(CH <sub>2</sub> ) <sub>5</sub> CH <sub>3</sub>	10.8	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>5</sub> O- 	11.4
Br-(CH <sub>2</sub> ) <sub>2</sub> - 	4.2	 -(CH <sub>2</sub> ) <sub>2</sub> O- 	4.5

Table A (continued)

<u>Alkyl halide or tosylate</u>	<u>mass</u> (g)	<u>Alkoxy aryl product</u>	<u>mass</u> (g)
 $\text{CH}_3$ -  - $\text{SO}_3^-(\text{CH}_2)_2\text{O}(\text{CH}_2)_6\text{CH}_3$	23.4	$\text{CH}_3(\text{CH}_2)_6\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_2\text{CH}_3$	20.9
 $\text{CH}_3$ -  - $\text{SO}_3^-(\text{CH}_2)_2\text{O}(\text{CH}_2)_7\text{CH}_3$	25.8	$\text{CH}_3(\text{CH}_2)_7\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_2\text{CH}_3$	7.9
 $\text{CH}_3$ -  - $\text{SO}_3^-(\text{CH}_2)_2\text{O}(\text{CH}_2)_9\text{CH}_3$	27.1	$\text{CH}_3(\text{CH}_2)_9\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_2\text{CH}_3$	21.0
$\text{I}-(\text{CH}_2)_3\text{CH}_3$	6.1	$\text{CH}_3(\text{CH}_2)_3\text{O}-\text{C}_6\text{H}_4-\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	12.3
$\text{I}-(\text{CH}_2)_5\text{CH}_3$	4.3	$\text{CH}_3(\text{CH}_2)_5\text{O}-\text{C}_6\text{H}_4-\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	4.7
$\text{I}-(\text{CH}_2)_2\text{CH}_3$	2.6	$\text{CH}_3(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	4.4
 $\text{CH}_3$ -  - $\text{SO}_3^-(\text{CH}_2)_2\text{O}(\text{CH}_2)_3\text{CH}_3$	2.7	$\text{CH}_3(\text{CH}_2)_3\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	2.6
 $\text{CH}_3$ -  - $\text{SO}_3^-(\text{CH}_2)_2\text{OC}(\text{CH}_3)_3$	2.7	$(\text{CH}_3)_3\text{CO}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	2.6
$\text{I}-(\text{CH}_2)_2\text{CH}_3$	3.8	$\text{CH}_3(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	1.4
 $\text{CH}_3$ -  - $\text{SO}_3^-(\text{CH}_2)_2\text{O}(\text{CH}_2)_3\text{CH}_3$	3.6	$\text{CH}_3(\text{CH}_2)_3\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	5.1
 $\text{CH}_3$ -  - $\text{SO}_3^-(\text{CH}_2)_2\text{OC}(\text{CH}_3)_3$	4.9	$(\text{CH}_3)_3\text{CO}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	5.2

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Preparation 2

The following nitrile and ester intermediates where R<sub>2</sub> is an alkoxy aryl moiety were prepared substantially in accordance with Procedure B, detailed above.

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Table B

Alkyl alcohol	mass (g)	Alkoxy Aryl Product	mass (g)
$\text{HO}-(\text{CH}_2)_2-\text{N}-(\text{CH}_2)_2\text{CH}_3$	3.6	$\text{CH}_3(\text{CH}_2)_2-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	6.2
$\text{HO}-(\text{CH}_2)_2-\text{N}-(\text{CH}_2)_2\text{CH}_2-\text{C}_6\text{H}_5$	6.1	$\text{C}_6\text{H}_5-\text{CH}_2-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	4.3
$\text{HO}-(\text{CH}_2)_2-\text{N}-(\text{CH}_2)_5\text{CH}_3$	0.5	$\text{CH}_3(\text{CH}_2)_5-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	0.8
$\text{HO}-(\text{CH}_2)_2-\text{N}-(\text{CH}_2)_2\text{CH}_2-\text{C}_6\text{H}_{10}$	0.5	$\text{C}_6\text{H}_{10}-\text{CH}_2-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	0.5
$\text{HO}-(\text{CH}_2)_2-\text{N}-(\text{CH}_2)_2\text{CH}_2-\text{C}_6\text{H}_4-\text{C}_6\text{H}_4-\text{C}_6\text{H}_5$	2.3	$\text{C}_6\text{H}_5-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	1.3
$\text{HO}-(\text{CH}_2)_2-\text{N}-(\text{CH}_2)_2\text{CH}_2-\text{C}_6\text{H}_{10}$	9.3	$\text{C}_6\text{H}_{10}-\text{CH}_2-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	9.6
$\text{HO}-\text{CH}_2-\text{C}_6\text{H}_4-(\text{CH}_2)_3\text{CH}_3$	10.0	$\text{CH}_3(\text{CH}_2)_3-\text{CH}_2\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_2\text{CH}_3$	13.6

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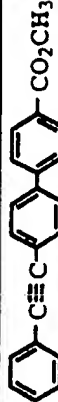


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Preparation 3

The following ester intermediates where R<sub>2</sub> is an alkynyl or alkenyl aryl moiety were prepared substantially in accordance with Procedure C, detailed above.



Table C

Alkene or alkyne	mass (g)	Aryl halide	mass (g)	Alkenyl or alkynyl aryl product	mass (g)
$\text{HC}\equiv\text{C}-(\text{CH}_2)_5\text{CH}_3$	12.1		28.8	$\text{CH}_3(\text{CH}_2)_5-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	26.2
$\text{HC}\equiv\text{C}-(\text{CH}_2)_5\text{CH}_3$	6.1		14.4	$\text{CH}_3(\text{CH}_2)_5-\text{CH}=\text{CH}^{(\text{trans})}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	0.6
$\text{HC}\equiv\text{C}-(\text{CH}_2)_7\text{CH}_3$	15.2		28.8	$\text{CH}_3(\text{CH}_2)_7-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	28.1
$\text{HC}\equiv\text{C}-\text{C}_6\text{H}_5$	1.9		5.1		1.9
$\text{HC}\equiv\text{C}-\text{Si}(\text{CH}_3)_3$	4.3		11.5	$(\text{CH}_3)_3\text{Si}-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	11.2
$\text{HC}\equiv\text{C}-\text{C}_6\text{H}_5$	1.8		6.0		2.6
$\text{HC}\equiv\text{C}-(\text{CH}_2)_3\text{CH}_3$	1.4		6.0	$\text{CH}_3(\text{CH}_2)_3-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	5.1
$\text{HC}\equiv\text{C}-\text{Si}(\text{CH}_3)_3$	10.9		40.0	$(\text{CH}_3)_3\text{Si}-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	23.3

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Table C (continued)

Alkene or alkyne	mass (g)	Aryl halide	mass (g)	Alkenyl or alkynyl aryl product	mass (g)
$\text{HC}\equiv\text{C}-(\text{CH}_2)_7\text{CH}_3$	7.6		11.3	$\text{CH}_3(\text{CH}_2)_7-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{O}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	11.4
$\text{HC}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	10.5		9.7		10.2
$\text{HC}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	22.2		34.4		19.4
$\text{HC}\equiv\text{C}-\text{C}_6\text{H}_4-\text{CO}_2\text{CH}_3$	1.2		1.2		1.5

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Preparation 4

The following ester intermediates where R<sub>2</sub> is a terphenyl moiety were prepared substantially in accordance with Procedure D, detailed above.

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Table D.1

















<u>Aryl halide</u> (R <sup>D</sup> is bromide)	<u>mass</u> (g)	<u>Boronic acid</u> <u>reactant</u> (R <sup>D</sup> is B(OH) <sub>2</sub> mass (g))
R <sup>D</sup> -  -O(CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	10.6	6.1
R <sup>D</sup> -  -O(CH <sub>2</sub> ) <sub>4</sub> CH <sub>3</sub>	31.0	12.0
R <sup>D</sup> -  -O(CH <sub>2</sub> ) <sub>5</sub> CH <sub>3</sub>	10.9	4.1
R <sup>D</sup> -  -O(CH <sub>2</sub> ) <sub>2</sub> -O(CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	13.6	5.7
R <sup>D</sup> -  -O(CH <sub>2</sub> ) <sub>2</sub> -OC(CH <sub>3</sub> ) <sub>3</sub>	5.0	1.9

Table D.2

Boronic acid reactant	mass (g)	$\text{CH}_3\text{CO}_2$ 	mass (g)	Product	mass (g)
$(\text{HO})_2\text{B}$  $\text{O}(\text{CH}_2)_3\text{CH}_3$	5.0	3.2		$\text{CH}_3\text{CO}_2$  $\text{O}(\text{CH}_2)_3\text{CH}_3$	4.2
$(\text{HO})_2\text{B}$  $\text{O}(\text{CH}_2)_4\text{CH}_3$	6.0	3.7		$\text{CH}_3\text{CO}_2$  $\text{O}(\text{CH}_2)_4\text{CH}_3$	5.2
$(\text{HO})_2\text{B}$  $\text{O}(\text{CH}_2)_5\text{CH}_3$	3.4	2.8		$\text{CH}_3\text{CO}_2$  $\text{O}(\text{CH}_2)_5\text{CH}_3$	3.5
$(\text{HO})_2\text{B}$  $\text{O}(\text{CH}_2)_2 - \text{O}(\text{CH}_2)_3\text{CH}_3$	3.7	3.6		$\text{CH}_3\text{CO}_2$  $\text{O}(\text{CH}_2)_2 - \text{O}(\text{CH}_2)_3\text{CH}_3$	3.7
$(\text{HO})_2\text{B}$  $\text{O}(\text{CH}_2)_2 - \text{OC}(\text{CH}_3)_3$	1.8	1.5		$\text{CH}_3\text{CO}_2$  $\text{O}(\text{CH}_2)_2 - \text{OC}(\text{CH}_3)_3$	2.2

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Preparation 5

5           The following activated esters were prepared  
substantially in accordance with Procedure G, detailed  
above.

Table G

Carboxylic acid	mass (g)	mass (g) activated ester (2,4,5-trichlorophenyl ester)
$\text{CH}_3(\text{CH}_2)_3\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	1.9	1.8
$\text{Cyclopentyl}-\text{CH}_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	4.2	4.4
$(\text{CH}_3)_2\text{CH}(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	3.0	1.7
$\text{CH}_3(\text{CH}_2)_4\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	2.2	1.3
$\text{CH}_3(\text{CH}_2)_4\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	5.7	5.1
$\text{Cyclopentyl}-\text{CH}_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	4.4	3.1
$(\text{CH}_3)_3\text{C}(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	2.3	2.6
$(\text{CH}_3\text{CH}_2)_2\text{CHCH}_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	1.5	0.8
$\text{CH}_3(\text{CH}_2)_5\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	5.3	4.8
$\text{Cyclohexyl}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	3.1	1.0
$\text{CH}_3(\text{CH}_2)_6\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	5.6	2.9
$\text{CH}_3(\text{CH}_2)_7\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	7.8	6.6

Table G (continued)

Carboxylic acid	mass (g)	mass (g) (2,4,5-trichlorophenyl ester)	mass (g) activated ester
$\text{CH}_3(\text{CH}_2)_9\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	6.4		1.3
$\text{CH}_3(\text{CH}_2)_3-\text{C}_6\text{H}_4-\text{CH}_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	4.0		3.2
$\text{CH}_3(\text{CH}_2)_3\text{O}-\text{C}_6\text{H}_4-\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	5.8		1.4
$\text{CH}_3(\text{CH}_2)_5\text{O}-\text{C}_6\text{H}_4-\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	3.8		2.4
$\text{CH}_3(\text{CH}_2)_3\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{COOH}$	2.9		2.5
$\text{CH}_3(\text{CH}_2)_3\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{COOH}$	2.0		1.5
$(\text{CH}_3)_3\text{CO}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{COOH}$	2.0		1.3
$\text{CH}_3(\text{CH}_2)_3\text{O}-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{COOH}$	6.5		5.2
$\text{CH}_3(\text{CH}_2)_3\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{COOH}$	4.9		5.2
$(\text{CH}_3)_3\text{CO}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{COOH}$	4.6		2.1

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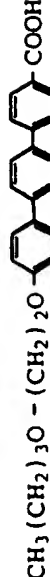
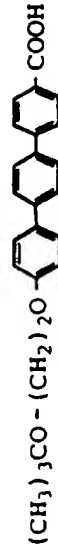
Table G (continued)

Carboxylic acid	mass (g)	mass (g) (2,4,5-trichlorophenyl ester)	activated ester
$\text{CH}_3(\text{CH}_2)_2-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	3.3	1.5	
$\text{C}_6\text{H}_5-\text{CH}_2-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	3.0	2.3	
$\text{CH}_3(\text{CH}_2)_5-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	1.0	1.0	
$\text{C}_6\text{H}_{11}-\text{CH}_2-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	2.0	0.8	
$\text{C}_6\text{H}_{11}-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	7.2	0.8	
$\text{C}_6\text{H}_{11}-\text{CH}_2-\text{N}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{COOH}$	7.5	7.3	
$\text{C}_6\text{H}_5-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{COOH}$	2.0	0.6	
$\text{CH}_3(\text{CH}_2)_3-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{COOH}$	1.1	0.6	
$\text{CH}_3(\text{CH}_2)_5-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{COOH}$	4.6	3.5	
$\text{CH}_3(\text{CH}_2)_5-\text{CH}=\text{C}(\text{trans})-\text{C}_6\text{H}_4-\text{COOH}$	1.2	0.5	
$\text{CH}_3(\text{CH}_2)_7-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{COOH}$	11.1	13.2	



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Table G (continued)

Carboxylic acid	mass (g)	mass activated ester (2,4,5-trichlorophenyl ester)	mass (g)
	1.5		1.5
	8.3		13.2
	0.8		1.2
	3.3		4.8
	3.0		2.5
	2.3		3.9
	3.3		4.4
	1.3		1.9

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Example 1N-Acylation of Cyclic Peptide Nuclei

5       The N-acyl cyclic peptide derivatives listed in  
Table 3, below were prepared by dissolving Echinocandin B  
(A-30912A) nucleus (compound of formula IB where R', R'',  
and R''' are each methyl, R<sup>x1</sup> and R<sup>x2</sup> are each hydroxy, R<sup>y1</sup>,  
R<sup>y2</sup>, R<sup>y3</sup>, and R<sup>y4</sup> are each hydroxy, and R is hydroxy), and  
10       the activated ester (2,4,5-trichlorophenol ester)  
intermediates, described in Preparation 6, in 25-50 ml of  
dimethylformamide. The resultant reaction mixture was  
stirred for approximately 17-65 hours at room temperature  
and then the solvent was removed in vacuo to provide a  
residue. This residue was slurried in ether, collected by  
15       filtration, washed with methylene chloride and then  
dissolved in methanol or a 1:1 (v/v) acetonitrile/water  
mixture. The resultant solution is subjected to reverse  
phase HPLC (C18; eluent of 20-40% aqueous acetonitrile  
containing 0.5% monobasic ammonium phosphate (w/v); 20  
20       ml/min.; 230 nm). After removing the unreacted A30912A  
nucleus, the desired product is eluted from the column  
using an eluent of aqueous acetonitrile. The fractions  
containing the desired product are combined and then  
concentrated in vacuo or lyophilized to provide the desired  
25       acylated nucleus. The product may be analyzed using  
reverse phase HPLC (C18; 40% aqueous acetonitrile  
containing 0.5% monobasic ammonium phosphate (w/v); 2  
ml/min; 230 nm) or using MS (FAB).

30       For example, the compound depicted in Table 3II,  
below, was prepared substantially according to this  
procedure, using 348.1 g (60.2 mmol) of the A30912A  
nucleus, 26.0 g (48.2 mmol) of the 2,4,5-trichlorophenol  
ester of [(4"-pentyloxy)-1,1':4',1"-terphenyl]-4-  
35       carboxylic acid in 8.5 liter of dimethylformamide. The  
resultant reaction mixture was allowed to react for  
approximately forty eight hours and then concentrated in

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vacuo and purified using HPLC to provide 18 g of compound 3II.

MS (FAB): 1140.5103 ( $M^+$ ).

5           Compounds A-PP (listed in Table 3 below) were prepared substantially as described above.

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Table 3

Ex. No.	R <sub>2</sub>	Ester (mg)	A30912A Nucleus (g)	Product (mg)	MS (FAB)	HPLC R <sub>T</sub> (min)
3A		561	1.0	235	1072*	4.08
3B		576	1.0	294	1062*	4.46
3C		579	1.0	355	1086*	5.75
3D		634	1.0	359	1130*	5.79
3E		289	0.5	81	1083*	6.08
3F		594	1.0	295	1098*	6.44
3G		596	1.0	270	1100*	8.15
3H		596	1.0	359	1100*	9.13
3I		596	1.0	301	1100*	10.24
3J		629	1.0	180	1104**	--
3K		287	0.5	110	1082*	4.52
3L		593	1.0	307	1096*	7.28

\* (m-1)+[Na]<sup>+</sup>; \*\* (m+1); \*\*\* m+[Na]<sup>+</sup>; † m+1; †† m+[Li]<sup>+</sup>.

Table 3 (continued)

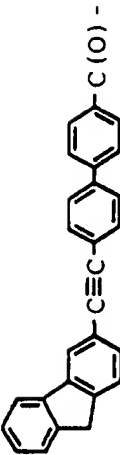
Ex. No.	R <sub>2</sub>	Estex (mg)	A30912A Nucleus (g)	Product (mg)	MS (FAB)	HPLC R <sub>T</sub> (min)
3M		313	0.5	104	1124*	19.04
3N		579	1.0	293	1032*	6.14
3O		291	0.5	98	1088*	3.96
3P		616	1.0	341	1116*	11.56
3Q		2400	3.2	3000	1194.5213†	--
3R		1300	1.5	2400	1194.5247†	--
3S		4600	7.4	1300	1126.5025†	--
3T		683	1.0	384	1147**	1.92
3U		1490	2.0	116	1195**	2.06
3V		1000	1.2	194	1190***	2.41
3W		734	0.9	303	1202*	2.21
3X		810	1.0	230	1187**	2.52

\* (m-1)+[Na]<sup>+</sup>; \*\* (m+1); \*\*\* m+[Na]<sup>+</sup>; † m+1; †† m+[Li]<sup>+</sup>.

Ex. No.	R <sub>2</sub>	Ester (mg)	A10912A Nucleus (g)	Product (mg)	MS (FAB)	HPLC R <sub>T</sub> (min)
3Y		750	1.0	126	1201**	3.50
3Z		596	1.0	190	1078**	6.30
3AA	$\text{CH}_3(\text{CH}_2)_3 - \text{C}\equiv\text{C} - \text{C}_6\text{H}_4 - \text{C}(\text{O}) -$	571	1.0	295	1058**	7.91
3BB	$\text{CH}_3(\text{CH}_2)_5 - \text{C}\equiv\text{C} - \text{C}_6\text{H}_4 - \text{C}(\text{O}) -$	511	1.0	322	1032*	5.10
3CC	$\text{CH}_3(\text{CH}_2)_5 - \text{CH}=\text{C}^{\text{(trans)}} - \text{C}_6\text{H}_4 - \text{C}(\text{O}) -$	514	1.0	287	1034*	6.14
3DD	$\text{CH}_3(\text{CH}_2)_7 - \text{C}\equiv\text{C} - \text{C}_6\text{H}_4 - \text{C}(\text{O}) -$	546	1.0	285	1060*	12.48
3EE		501	1.0	218	1002**	2.53
3FF	$\text{CH}_3(\text{CH}_2)_7 - \text{C}\equiv\text{C} - \text{C}_6\text{H}_4 - \text{C}(\text{O}) -$	534	1.0	215	1050**	7.59
3GG		566	1.0	81	1054**	3.89
3HH	$\text{CH}_3(\text{CH}_2)_3\text{O} - \text{C}_6\text{H}_4 - \text{C}_6\text{H}_4 - \text{C}(\text{O}) -$	4600	7.4	1300	1126.5025†	--
3II	$\text{CH}_3(\text{CH}_2)_4\text{O} - \text{C}_6\text{H}_4 - \text{C}_6\text{H}_4 - \text{C}(\text{O}) -$	2500	3.7	5100	1140.5103†	--
3JJ	$\text{CH}_3(\text{CH}_2)_5\text{O} - \text{C}_6\text{H}_4 - \text{C}_6\text{H}_4 - \text{C}(\text{O}) -$	3500	5.0	1400	--	--

$$* \quad (m-1) + [\text{Na}]^+; \quad ** \quad (m+1); \quad *** \quad m + [\text{Na}]^+; \quad \dagger \quad m+1; \quad \dagger\dagger \quad m + [\text{Li}]^+.$$

Table 3 (continued)

Ex- No.	R <sub>2</sub>	Ester (mg)	A30912A Nucleus (g)	Product (mg)	MS (FAB)	HPLC RT (min)
3KK	$\text{CH}_3(\text{CH}_2)_3\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}_6\text{H}_4-\text{C}(\text{O})-$	4400	6.7	6500	1170.5234†	--
3LL	$(\text{CH}_3)_3\text{CO}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}_6\text{H}_4-\text{C}(\text{O})-$	1900	2.9	1400	1170.5261†	--
3MM	$\text{CH}_3(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{C}(\text{O})-$	5200	6.9	1400	1142.4951†	--
3NN	$(\text{CH}_3)_3\text{CO}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{C}(\text{O})-$	2100	2.5	2000	1200.5336†	--
3OO	$\text{CH}_3(\text{CH}_2)_3\text{O}-(\text{CH}_2)_2\text{O}-\text{C}_6\text{H}_4-\text{C}\equiv\text{C}-\text{C}_6\text{H}_4-\text{C}(\text{O})-$	5200	6.4	1100	1194.5282†	--
3PP		1800	2.6	200	1166.4758†	--

\* (m-1)+[Na]<sup>+</sup>; \*\* (m+1); \*\*\* m+[Na]<sup>+</sup>; † m+1; †† m+[Li]<sup>+</sup>.

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Example 2Dideoxycilofungin

To a suspension of 10.00 g (9.71 mmol) of  
5 cilofungin in 100 ml of methylene chloride, was added a  
solution of 96 ml (602 mmol) of triethylsilane in 50 ml of  
methylene chloride. Then, a solution of 46.4 ml (602 mmol)  
of trifluoroacetic acid in 50 ml of methylene chloride was  
slowly added, over 15 minutes. The resultant solution was  
10 stirred at room temperature for approximately two hours and  
then concentrated *in vacuo* to provide a residue. This  
residue was triturated with diethyl ether and then purified  
using reverse phase HPLC (C18; gradient eluent of 10-20%  
acetonitrile in water (v/v); 500 psi). The fractions  
15 containing the desired compound were combined, concentrated  
*in vacuo*, and then lyophilized from p-dioxane to provide  
the desired titled compound.

Yield: 6.66 g (68.7%).

MS (FAB) for  $C_{49}H_{72}N_7O_{15}$ 

20 Calc. 998.5086;

Found: 998.512.

UV:  $\lambda$ (EtOH)nm( $\epsilon$ ) 202.60(61012), 256.20(18569).Example 3

25

Preparation of the dideoxy compound where R', R'', and R'''  
are each methyl, R<sup>x1</sup> and R<sup>x2</sup> are each hydroxy, R<sup>y1</sup>, R<sup>y2</sup>,  
R<sup>y3</sup>, R<sup>y4</sup> are each hydroxy, R is hydroxy and R<sub>2</sub> is the acyl  
group depicted in Table 3II

30

To a mixture of 5 g (4.4 mmol) of the compound  
of Table 3II and 17 ml of trifluoroacetic acid in 250 ml of  
methylene chloride, was added 35 ml of triethylsilane.  
When the reaction was substantially complete, as indicated  
by HPLC (C18, eluent of 55% acetonitrile; 2 ml/min; 280 nm;  
35 R<sub>T</sub> (starting material) = 4.19 min.; R<sub>T</sub> (product) = 6.40  
min.), the reaction mixture was concentrated *in vacuo* to  
provide a solid. This solid was slurried in 100 ml of 50%  
aqueous acetone and then dissolved by adjusting the pH of



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the mixture to approximately pH 7. The resultant solution was poured into a large volume of water (approximately 1 liter) resulting in the precipitation of a white solid. This solid was isolated by filtration through a sintered glass funnel, washed with diethyl ether and then dried in vacuo at 55°C to provide 3.718 g of the titled compound. The funnel was washed with methanol to collect the remaining solid, which was dried in vacuo to provide an additional 0.154 g of the titled compound.

Yield: 3.872 g (79%).

MS (FAB): m/e 1108.7 (M)

HPLC: (eluent of 55% acetonitrile; 2 ml/min.; 280 nm):  
R<sub>T</sub> = 6.43 min.

15

Example 4Preparation of dideoxy cyclic hexapeptides

The following dideoxy compounds were prepared substantially in accordance with the procedure detailed in Example 3 using the designated amounts of a compound of formula IC where R', R'' and R''' are methyl, R<sup>x1</sup> and R<sup>x2</sup> are hydroxy, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup> and R<sup>y4</sup> are hydroxy, R is hydroxy and R<sub>2</sub> is as designated, triethylsilane (TES) and trifluoroacetic acid (TFA).

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Table 4

Ex. No.	R <sub>2</sub>	formula IC (g)	TES (g)	TF		Yield (g)
				A	(g)	
A		0.500	0.256	0.251		0.095
B		0.500	2.47	2.42		0.063
C		0.500	2.63	2.57		0.392
D		2.00	9.49	9.72		1.47
E		0.500	3.50	3.44		0.291

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Example 5

5     Preparation of the compound where R', R'', and R''' are each methyl, R<sup>x1</sup> and R<sup>x2</sup> are each hydroxy, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup>, R<sup>y4</sup> are each hydroxy, R<sub>0</sub> is ethylphosphonate and R<sub>2</sub> is the acyl group depicted in Table 3II

A. Protection

10             To a solution of 2.00 g (1.75 mmol) of the compound of Table 3II in 50 ml of dioxane at room temperature, was added 25 ml (175 mmol) of 2-(trimethylsilyl)ethanol and p-toluenesulfonic acid (15  
15     mol percent). The resultant reaction mixture was allowed to react at room temperature for approximately three hours. When the reaction was substantially complete, as indicated by HPLC, the reaction was quenched with solid sodium bicarbonate and filtered. The desired titled compound was isolated from the filtrate using reverse phase HPLC (50%  
20     acetonitrile/50% water; 50 ml/min.; 280 nm).  
Yield: 807 mg.

B. Formation of ethyl phosphonate derivative

25             To a cold (0°C) solution of 234.1 mg (0.191 mmol) of the titled compound of Example 5A in 10 ml of tetrahydrofuran, was added 0.21 ml (0.210 mmol) of a 1.0M solution of lithium bis(trimethylsilyl)amide (LHMDS) in hexanes, dropwise. The resultant mixture was allowed to stir for approximately twenty minutes followed by the  
30     dropwise addition of 24.5 µl (0.223 mmol) of ethylphosphonic dichloride. The reaction mixture was stirred for approximately thirty minutes, quenched with 1 ml of water and then concentrated in vacuo to provide a white solid.  
35     Yield: 42 mg.

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HPLC (50% acetonitrile/50% water; 50 ml/min.; 280 nm):  
R<sub>T</sub>=1.37

C. Deprotection

5                   To a mixture of 40.0 mg (0.028 mmol) of the  
                  titled compound of Example 5B in 20 ml of methylene  
                  chloride, was added 35  $\mu$ l (0.28 mmol) of boron trifluoride  
                  etherate, dropwise. The resultant reaction mixture was  
                  allowed to react for approximately thirty minutes and then  
10                  was quenched with 1.0 ml of water, resulting in the  
                  formation of a white precipitate. The reaction mixture was  
                  trituated with diethyl ether and then filtered to provide  
                  a light yellow solid.  
                  Yield: 12 mg.  
15                  MS (FAB): 1238.6 (M+Li).

Example 6

20                  Preparation of the compound where R', R'', and R''' are each  
                  methyl, RX1 and RX2 are each hydroxy, RV1, RV2, RV3, RV4 are  
                  each hydroxy, R0 is methylphosphonate and R2 is the acyl  
                  group depicted in Table 3II

A. Formation of methyl phosphonate derivative

25                  The subtitled compound was prepared  
                  substantially in accordance with the procedure detailed in  
                  Example 5B, above using using 271.1 mg (0.221 mmol) of the  
                  titled compound of Example 5A, 0.24 ml (0.24 mmol) of a  
                  1.0M solution of LHMDS in hexanes and 35.3 mg (0.266 mmol  
30                  of methylphosphonic dichloride in 10 ml of tetrahydrofuran  
                  to provide 40 mg of crude material that was used without  
                  further purification.

B. Deprotection

35                  The subtitled compound was prepared  
                  substantially in accordance with the procedure detailed in  
                  Example 5C, using the compound isolated in Example 6A and

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35  $\mu$ l of (0.28 mmol) of boron trifluoride etherate to provide a white/gray solid.  
MS (FAB): 1200.5 (M-H<sub>2</sub>O).

5

Example 7

Preparation of the compound where R', R'', and R''' are each methyl, R<sup>x1</sup> and R<sup>x2</sup> are each hydroxy, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup>, R<sup>y4</sup> are each hydroxy, R<sub>0</sub> is phenylphosphonate and R<sub>2</sub> is the acyl group depicted in Table 3II

10

A. Formation of phenyl phosphonate derivative

The subtitled compound was prepared substantially in accordance with the procedure detailed in Example 5B, above, using using 359.6 mg (0.294 mmol) of the titled compound of Example 5A, 0.333 ml (0.323 mmol) of a 1.0M solution of LHMDs in hexanes and 50  $\mu$ l of phenylphosphonic dichloride to provide 52 mg of crude material which was used without further purification.

15

B. Deprotection

The subtitled compound was prepared substantially in accordance with the procedure detailed in Example 5C, using the compound isolated in Example 6A and 361  $\mu$ l of boron trifluoride etherate to provide a yellowish solid.

20

Yield: 32 mg.

MS (FAB): 1262.4 (M-H<sub>2</sub>O).

25

Example 8

Preparation of the compound where R', R'', and R''' are each methyl, R<sup>x1</sup> and R<sup>x2</sup> are each hydrogen, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup>, R<sup>y4</sup> are each hydroxy, R<sub>0</sub> is isopropylphosphate and R<sub>2</sub> is the acyl group depicted in Table 3II

30

A. Isopropylidichlorophosphate

To a solution of 1.55 ml (16.6 mmol) of phosphorous oxychloride in 5 ml of carbon tetrachloride,

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was added 1.28 ml (16.6 mmol) of isopropanol, under nitrogen, resulting in an increase in temperature. An ice bath was used as necessary to keep the temperature between 20°C and 35°C. The reaction mixture was allowed to react for approximately seven hours at room temperature, under nitrogen. The resultant mixture was concentrated in vacuo to provide a clear oil.

Yield: 1.9 g (65%).

10 B. Formation of isopropyl phosphate derivative

To a cold (0°C) solution containing 0.5 g (0.45 mmol) of the titled compound of Example 3 in 10 ml of tetrahydrofuran and 54 µl (0.54 mmol) of lithium trimethylsilanolate (LiOTMS), was added 88 mg (0.5 mmol) of the subtitled compound of Example 8A. The resultant reaction mixture was stirred for approximately ten minutes. Additional LiOTMS was added to the reaction mixture until the pH of the mixture was basic. When the reaction was substantially complete, as indicated by HPLC, the reaction was quenched with water, stirred for approximately one hour and then concentrated in vacuo to provide a yellow solid. This solid was purified using HPLC (eluent of 45% acetonitrile/45% water/10% trifluoroacetic acid (1% aqueous solution)) to provide a white solid.

25 Yield: 105 mg.

MS (FAB): 1230.4 (M<sup>+</sup>)

Example 9

30 Preparation of the compound where R', R'', and R''' are each methyl, R<sup>x1</sup> and R<sup>x2</sup> are each hydrogen, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup>, R<sup>y4</sup> are each hydroxy, R<sub>0</sub> is butylphosphate and R<sub>2</sub> is the acyl group depicted in Table III

35 A. Butyldichlorophosphate

The desired subtitled compound was prepared substantially in accordance with the procedure detailed in Example 8A, using 1.25 ml (13.5 mmol) of phosphorous

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oxychloride, 1 g (13.5 mmol) of butanol in 5 ml of carbon tetrachloride to provide a colorless oil.

Yield: 2.3 g (89%).

5     B. Formation of butyl phosphate derivative

          The desired subtitled compound was prepared substantially in accordance with the procedure detailed in Example 8B, using 0.5 g (0.45 mmol) of the titled compound of Example 3, LiOTMS and 95 mg (0.50 mmol) of the subtitled  
10    compound of Example 9A to provide a yellow solid. This solid was purified using reverse phase HPLC (gradient eluent of 45% acetonitrile/45% water/10% (1% aqueous) trifluoroacetic acid → 50% acetonitrile/40% water/10% (1% aqueous) trifluoroacetic acid) to provide a 126 mg of the  
15    desired compound.

MS (FAB): 1244.4 (M<sup>+</sup>)

Example 10

20    Preparation of the compound where R', R'', and R''' are each methyl, R<sup>x1</sup> and R<sup>x2</sup> are each hydrogen, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup>, R<sup>y4</sup> are each hydroxy, R<sub>0</sub> is methylphosphate and R<sub>2</sub> is the acyl group depicted in Table 3II

          To a mixture of 500 mg (0.45 mmol) of the titled  
25    compound of Example 3 and 0.5 mL (0.5 mmol) of LiOTMS in 5 ml of tetrahydrofuran, was added 0.075 ml (0.75 mmol) of methyl dichlorophosphate resulting in the dissolution of solid material. The reaction was monitored by HPLC (eluent of 70% acetonitrile; 2 ml/min.; 280 nm) resulting in the  
30    addition of an additional 0.7 ml of LiOTMS and 0.02 ml of methyl dichlorophosphate to the reaction mixture. When the reaction was substantially complete, as indicated by HPLC (eluent of 50% acetonitrile containing 0.1% trifluoroacetic acid; 2 ml/min.; 280 nm), the desired compound was isolated  
35    using HPLC (eluent of 40% acetonitrile containing 0.1% trifluoroacetic acid; 90 ml/min.; 280 nm). The fractions containing the desired compound were combined and concentrated in vacuo to provide 232 mg of the titled

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compound (81% pure by HPLC). This compound was purified using HPLC (step gradient eluent of 30% → 35% → 40% acetonitrile containing 0.1% trifluoroacetic acid; 90 ml/min.; 280 nm) to provide 109 mg of the titled compound (94% pure).

5 MS (FAB): m/e 1202.6 (M<sup>+</sup>).

#### Example 11

10 Preparation of the compound where R', R'', and R''' are each methyl, R<sup>x1</sup> and R<sup>x2</sup> are each hydrogen, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup>, R<sup>y4</sup> are each hydroxy, R<sub>0</sub> is hexylphosphonate and R<sub>2</sub> is the acyl group depicted in Table 3II

15 To a cold (0°C) mixture of 1 g (0.902 mmol) of the titled compound of Example 3 in 5 ml of tetrahydrofuran, was added 1.35 ml of a 1M solution of LHMDS in tetrahydrofuran (1.35 mmol), dropwise. After stirring the resulting mixture for approximately thirty minutes, 309 µl (1.804 mmol) of hexyldichlorophosphate was

20 added and the reaction mixture was allowed to warm to room temperature, followed by the addition of water. The resultant reaction mixture was reduced to dryness in vacuo to provide the desired titled compound.

Yield: 102 mg.

25 MS (FAB): Calcd: 1262.5978 (M<sup>+</sup>Li);  
Found: 1262.5979 (M<sup>+</sup>Li).

#### Example 12

30 Preparation of the compound where R', R'', and R''' are each methyl, R<sup>x1</sup> and R<sup>x2</sup> are each hydrogen, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup>, R<sup>y4</sup> are each hydroxy, R<sub>0</sub> is methylphosphonate and R<sub>2</sub> is the acyl group depicted in Table 3II

35 The titled compound was prepared substantially in accordance with the procedure detailed in Example 11, using 221.9 mg (0.200 mmol) of the titled compound of Example 3 and 0.240 ml of a 1M solution of LHMDS in hexanes (0.240 mmol) and 35 mg (0.26 mmol) of methyl phosphonic dichloride in 20 ml of tetrahydrofuran.



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Yield: 44 mg.

MS (FAB): 1192.2 (M+Li)

Example 13

Preparation of the compound where R', R'', and R''' are each methyl, R<sup>x1</sup> and R<sup>x2</sup> are each hydroxy, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup>, R<sup>y4</sup> are each hydroxy, R<sub>0</sub> is methylphosphate and R<sub>2</sub> is the acyl group depicted in Table 3II

A. Formation of methyl phosphate derivative

To a cold (0°C) mixture of 400 mg (0.32 mmol) of the subtitled compound of Example 5A and 0.36 mL (0.36 mmol) of LiOTMS (1M solution in methylene chloride) in 5 ml of tetrahydrofuran, under nitrogen, was added 0.04 ml (0.4 mmol) of methyl dichlorophosphate. When the reaction was substantially complete, as indicated by HPLC (eluent of 80% acetonitrile; 2 ml/min.; 280 nm), several aliquots of lithium hydroxide were added to the mixture. The desired compound was isolated using HPLC (eluent of 60% acetonitrile containing 0.1% trifluoroacetic acid; 90 ml/min.; 280 nm). The fractions containing the desired compound were combined and concentrated *in vacuo* to provide 129.8 mg of the subtitled compound.

Yield: 30%.

HPLC (eluent of 65% acetonitrile containing 0.1% trifluoroacetic acid; 2 ml/min.; 280 nm): R<sub>T</sub>=4.28 min.

B. Deprotection

To a cold (0°C) mixture of 118 mg (0.09 mmol) of the subtitled compound of Example 13A in 3 ml of methylene chloride, was added 35 µl (0.28 mmol) of boron trifluoride etherate. The resultant reaction mixture was allowed to react for approximately ten minutes and then was quenched with several drops of water, resulting in the formation of a white precipitate. The reaction mixture was concentrated *in vacuo* to provide a residue. This residue was slurried in diethyl ether and then filtered to provide a solid which

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was dried in vacuo. The resultant product was determined to be 92% pure using HPLC (eluent of 50% acetonitrile containing 1% trifluoroacetic acid; 2 ml/min.; 280 nm;  $R_T=3.92$  min)

5 Yield: 88 mg (80%).  
MS (FAB): 1216.4 (M-H<sub>2</sub>O)  
1256.3 (M+Na).

#### Example 14

10

Preparation of the compound where R', R'', and R''' are each methyl, RX1 and RX2 are each hydroxy, RV1, RV2, RV3, RV4 are each hydroxy, R<sub>0</sub> is ethylphosphate and R<sub>2</sub> is the acyl group depicted in Table 3II

15

#### A. Formation of ethyl phosphate derivative

To a cold (0°C) mixture of 400 mg (0.32 mmol) of the subtitled compound of Example 5A and 0.36 mL (0.36 mmol) of LiOTMS (1M solution in methylene chloride) in 5 ml of tetrahydrofuran, under nitrogen, was added 0.47 ml (0.4 mmol) of ethyl dichlorophosphate. When the reaction was substantially complete, as indicated by HPLC (eluent of 80% acetonitrile; 2 ml/min.; 280 nm), approximately 0.5 ml of water was added to the mixture, dropwise. The desired compound was isolated using HPLC (eluent of 60% acetonitrile containing 0.1% trifluoroacetic acid; 90 ml/min.; 280 nm). The fractions containing the desired compound were combined and concentrated in vacuo to provide a residue. This residue was slurried in diethyl ether and then filtered to provide 67.8 mg of a solid. This resultant product was determined to be 71% pure using HPLC (eluent of 65% acetonitrile containing 0.1% trifluoroacetic acid; 2 ml/min.; 280 nm;  $R_T=5.99$  min.).

30

#### B. Deprotection

The desired titled compound was prepared substantially in accordance with the procedure detailed in Example 13B, using 67.8 mg of the subtitled compound of

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Example 14A and 0.1 ml (0.81 mmol) of the boron trifluoride etherate in methylene chloride. The resultant product was determined to be 89% pure using HPLC (eluent of 50% acetonitrile containing 0.1% trifluoroacetic acid; 2 ml/min.; 280 nm;  $R_T$ =5.85 min.)  
Yield: 51 mg.  
MS (FAB): 1230.3 (M-H<sub>2</sub>O).

The compounds of formula I have improved properties over the previously known N-acyl cyclic peptide antifungal compounds. For example, the present compounds have increased oral bioavailability, an important property for a systemic antifungal compound. In addition, the present compounds have enhanced antifungal activity and enhanced water solubility, relative to previously known compounds.

The compounds of formula I exhibit antifungal and antiparasitic activity. For example, the compounds of formula I inhibit the growth of various infectious fungi including Candida spp. such as C. albicans, C. parapsilosis, C. krusei, C. glabrata, or C. tropicalis, C. lusitaniae; Torulopus spp. such as T. glabrata; Aspergillus spp. such as A. fumigatus; Histoplasma spp. such as H. capsulatum; Cryptococcus spp. such as C. neoformans; Blastomyces spp. such as B. dermatitidis; Fusarium spp., Trichophyton spp., Pseudallescheria boydii, Coccidioides immitis, Sporothrix schenckii and the like.

Antifungal activity of a test compound is determined *in vitro* by obtaining the minimum inhibitory concentration (MIC) of the compound using a standard agar dilution test or a disc-diffusion test. The compound is then tested *in vivo* (in mice) to determine the effective dose of the test compound for controlling a systemic fungal infection.

Accordingly, the following compounds were tested for antifungal activity against C. albicans.

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Table 5  
Minimal inhibitory concentration against C. albicans

	<u>Example No.</u>	<u>MIC (<math>\mu</math>g/ml)</u>
5	5C	0.312
	6B	1.25
	7B	2.5
	8B	>80
	9B	>80
10	10	0.312
	11	1.25
	12	0.039
	13B	0.625
	14B	0.625

In addition, the effective dose of the following compounds for controlling a systemic fungal infection (C. albicans) was tested in vivo (mice).

Table 5  
ED<sub>50</sub> (mouse)

	<u>Example No.</u>	<u>ED<sub>50</sub> (mg/kg)</u>
25	5C	1.25
	6B	1.58
	7B	>2.5
	8B	1.02
	9B	0.39
30	10	0.47
	11	0.312
	12	0.79
	13B	1.86
	14B	1.38

The compounds of the invention also inhibit the growth of certain organisms primarily responsible for opportunistic infections in immunosuppressed individuals.

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For example the compounds of the invention inhibit the growth of Pneumocystis carinii the causative organism of pneumocystis pneumonia (PCP) in AIDS and other immunocompromised patients. Other protozoans that are inhibited by compounds of formula I include Plasmodium spp., Leishmania spp., Trypanosoma spp., Cryptosporidium spp., Isospora spp., Cyclospora spp., Trichomonas spp., Microsporidiosis spp. and the like.

The compounds of formula I are active in vitro and in vivo and are useful in combating either systemic fungal infections or fungal skin infections. Accordingly, the present invention provides a method of inhibiting fungal activity comprising contacting a compound of formula I, or a pharmaceutically acceptable salt thereof, with a fungus. A preferred method includes inhibiting *Candida albicans* or *Aspergillus fumigatis* activity. The present invention further provides a method of treating a fungal infection which comprises administering an effective amount of a compound of formula I, or a pharmaceutically acceptable salt thereof, to a host in need of such treatment. A preferred method includes treating a *Candida albicans* or *Aspergillus fumigatis* infection.

With respect to antifungal activity, the term "effective amount," means an amount of a compound of the present invention which is capable of inhibiting fungal activity. The dose administered will vary depending on such factors as the nature and severity of the infection, the age and general health of the host and the tolerance of the host to the antifungal agent. The particular dose regimen likewise may vary according to such factors and may be given in a single daily dose or in multiple doses during the day. The regimen may last from about 2-3 days to about 2-3 weeks or longer. A typical daily dose (administered in single or divided doses) will contain a dosage level of from about 0.01 mg/kg to about 100 mg/kg of body weight of an active compound of this invention. Preferred daily

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doses generally will be from about 0.1 mg/kg to about 60 mg/kg and ideally from about 2.5 mg/kg to about 40 mg/kg.

The present invention also provides pharmaceutical formulations useful for administering the antifungal compounds of the invention. Accordingly, the present invention also provides a pharmaceutical formulation comprising one or more pharmaceutically acceptable carriers, diluents or excipients and a compound of claim 1. The active ingredient in such formulations comprises from 0.1% to 99.9% by weight of the formulation, more generally from about 10% to about 30% by weight. By "pharmaceutically acceptable" it is meant that the carrier, diluent or excipient is compatible with the other ingredients of the formulation and not deleterious to the recipient thereof.

A compound of formula I may be administered parenterally, for example using intramuscular, subcutaneous, or intra-peritoneal injection, nasal, or oral means. In addition to these methods of administration, a compound of formula I may be applied topically for skin infections.

For parenteral administration the formulation comprises a compound of formula I and a physiologically acceptable diluent such as deionized water, physiological saline, 5% dextrose and other commonly used diluents. The formulation may contain a solubilizing agent such as a polyethylene glycol or polypropylene glycol or other known solubilizing agent. Such formulations may be made up in sterile vials containing the antifungal and excipient in a dry powder or lyophilized powder form. Prior to use, a physiologically acceptable diluent is added and the solution withdrawn via syringe for administration to the patient.

The present pharmaceutical formulations are prepared by known procedures using known and readily available ingredients. In making the compositions of the present invention, the active ingredient will generally be

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admixed with a carrier, or diluted by a carrier, or enclosed within a carrier which may be in the form of a capsule, sachet, paper or other container. When the carrier serves as a diluent, it may be a solid, semi-solid or liquid material which acts as a vehicle, excipient or medium for the active ingredient. Thus, the compositions can be in the form of tablets, pills, powders, lozenges, sachets, cachets, elixirs, suspensions, emulsions, solutions, syrups, aerosols, (as a solid or in a liquid medium), ointments containing, for example, up to 10% by weight of the active compound, soft and hard gelatin capsules, suppositories, sterile injectable solutions, sterile packaged powders and the like.

For oral administration, the antifungal compound is filled into gelatin capsules or formed into tablets. Such tablets may also contain a binding agent, a dispersant or other suitable excipients suitable for preparing a proper size tablet for the dosage and particular antifungal compound of the formula I. For pediatric or geriatric use the antifungal compound may be formulated into a flavored liquid suspension, solution or emulsion. A preferred oral formulation is linoleic acid, cremophor RH-60 and water and preferably in the amount (by volume) of 8% linoleic acid, 5% cremophor RH-60, 87% sterile water and a compound of formula I in an amount of from about 2.5 to about 40 mg/ml.

For topical use the antifungal compound may be formulated with a dry powder for application to the skin surface or it may be formulated in a liquid formulation comprising a solubilizing aqueous liquid or non-aqueous liquid, e.g., an alcohol or glycol.

The following formulation examples are illustrative only and are not intended to limit the scope of the invention in any way. The term "active ingredient" means a compound according to formula I or a pharmaceutically acceptable salt thereof.

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Formulation 1

Hard gelatin capsules are prepared using the following ingredients:

5		Quantity
		<u>(mg/capsule)</u>
	Active ingredient	250
	Starch, dried	200
10	Magnesium stearate	<u>10</u>
	Total	460 mg

Formulation 2

15      A tablet is prepared using the ingredients below:

		Quantity
		<u>(mg/capsule)</u>
20	Active ingredient	250
	Cellulose, microcrystalline	400
	Silicon dioxide, fumed	10
	Stearic acid	<u>5</u>
	Total	665 mg

25

The components are blended and compressed to form tablets each weighing 665 mg.

Formulation 3

30      An aerosol solution is prepared containing the following components:

		<u>Weight</u>
35	Active ingredient	0.25
	Methanol	25.75
	Propellant 22	



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(Chlorodifluoromethane)	<u>74.00</u>
Total	100.00

5       The active compound is mixed with ethanol and  
the mixture added to a portion of the propellant 22, cooled  
to  
-30°C and transferred to a filling device. The required  
amount is then fed to a stainless steel container and  
diluted with the remainder of the propellant. The valve  
10       units are then fitted to the container.

#### Formulation 4

Tablets, each containing 60 mg of active  
ingredient, are made as follows:

15       Active ingredient	60 mg
Starch	45 mg
Microcrystalline cellulose	35 mg
Polyvinylpyrrolidone	
20       (as 10% solution in water)	4 mg
Sodium carboxymethyl starch	4.5 mg
Magnesium stearate	0.5 mg
Talc	<u>1 mg</u>
Total	150 mg

25       The active ingredient, starch and cellulose are  
passed through a No. 45 mesh U.S. sieve and mixed  
thoroughly. The aqueous solution containing polyvinyl-  
pyrrolidone is mixed with the resultant powder, and the  
30       mixture then is passed through a No. 14 mesh U.S. sieve.  
The granules so produced are dried at 50°C and passed  
through a No. 18 mesh U.S. sieve. The sodium carboxymethyl  
starch, magnesium stearate and talc, previously passed  
through a No. 60 mesh U.S. sieve, are then added to the  
35       granules which, after mixing, are compressed on a tablet  
machine to yield tablets each weighing 150 mg.

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Formulation 5

Capsules, each containing 80 mg of active ingredient, are made as follows:

5	Active ingredient	80 mg
	Starch	59 mg
	Microcrystalline cellulose	59 mg
	Magnesium stearate	<u>2 mg</u>
	Total	200 mg

10           The active ingredient, cellulose, starch and magnesium stearate are blended, passed through a No. 45 mesh U.S. sieve, and filled into hard gelatin capsules in 200 mg quantities.

15

Formulation 6

Suppositories, each containing 225 mg of active ingredient, are made as follows:

	Active ingredient	225 mg
20	Saturated fatty acid glycerides	<u>2,000 mg</u>
	Total	2,225 mg

25           The active ingredient is passed through a No. 60 mesh U.S. sieve and suspended in the saturated fatty acid glycerides previously melted using the minimum heat necessary. The mixture is then poured into a suppository mold of nominal 2 g capacity and allowed to cool.

Formulation 7

30           Suspensions, each containing 50 mg of active ingredient per 5 ml dose, are made as follows:

	Active ingredient	50 mg
	Sodium carboxymethyl cellulose	50 mg
35	Syrup	1.25 ml
	Benzoic acid solution	0.10 ml
	Flavor	q.v.

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Color	q.v.
Purified water to total	5 ml

5 The active ingredient is passed through a No. 45 mesh U.S. sieve and mixed with the sodium carboxymethyl cellulose and syrup to form a smooth paste. The benzoic acid solution, flavor and color are diluted with a portion of the water and added, with stirring. Sufficient water is then added to produce the required volume.

10

#### Formulation 8

An intravenous formulation may be prepared as follows:

15

Active ingredient	100 mg
Isotonic saline	1,000 ml

20 The solution of the above ingredients generally is administered intravenously to a subject at a rate of 1 ml per minute.

The present invention further provides a method for treating or preventing the onset of Pneumocystis pneumonia in a host susceptible to Pneumocystis pneumonia which comprises administering an effective amount of a compound of formula I, or a pharmaceutically acceptable salt thereof, to a host in need of such treatment. The compounds of formula I can be used prophylactically to prevent the onset of the infection which is caused by the organism Pneumocystis carinii, or alternatively they can be used to treat a host that has been infected with Pneumocystis carinii. A compound of formula I may be administered parenterally, for example using intramuscular, intravenous or intra-peritoneal injection, orally or by inhaling directly into the airways of the lungs. A preferred mode of administration is inhalation of an aerosol spray formulation of a compound of formula I.

35

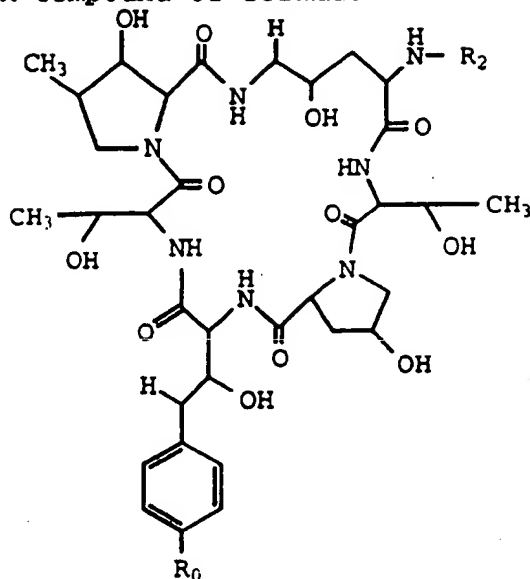
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With respect to antiparasitic activity, the term "effective amount," means an amount of a compound of the present invention which is capable of inhibiting parasitic activity. An effective amount of the compound of formula I is from about 3 mg/kg of patient body weight to about 100 mg/kg. The amount administered may be in a single daily dose or multiple doses of, for example, two, three or four times daily throughout the treatment regimen. The amount of the individual doses, the route of delivery, the frequency of dosing and the term of therapy will vary according to such factors as the intensity and extent of infection, the age and general health of the patient, the response of the patient to therapy and how well the patient tolerates the drug. It is known that Pneumocystis pneumonia infections in AIDS patients are highly refractory owing to the nature of the infection. For example, in severe, advanced infections the luminal surface of the air passages becomes clogged with infectious matter and extensive parasite development occurs in lung tissue. A patient with an advanced infection will accordingly require higher doses for longer periods of time. In contrast, immune deficient patients who are not severely infected and who are susceptible to Pneumocystis pneumonia can be treated with lower and less frequent prophylactic doses.

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## CLAIMS

1. A compound of formula I:



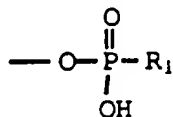
wherein:

R' is hydrogen, methyl or  $\text{NH}_2\text{C}(\text{O})\text{CH}_2-$ ;

R'' and R''' are independently methyl or hydrogen;

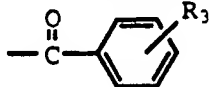
R<sup>x1</sup>, R<sup>x2</sup>, R<sup>y1</sup>, R<sup>y2</sup>, R<sup>y3</sup>, and R<sup>y4</sup> are independently hydroxy or hydrogen;

R<sub>0</sub> is a group of the formula



R<sub>1</sub> is C<sub>1</sub>-C<sub>6</sub> alkyl, C<sub>1</sub>-C<sub>6</sub> alkoxy, phenyl, p-halo-phenyl, p-nitrophenyl, phenoxy, benzyl, p-halo-benzyl, or p-nitro-benzyl;

- I) R<sub>2</sub> is a group of the formula



where:

A) R<sub>3</sub> is C<sub>1</sub>-C<sub>12</sub> alkyl, C<sub>1</sub>-C<sub>6</sub> alkoxy or quinolyl;

B) R<sub>3</sub> is  $-\text{O}-(\text{CH}_2)_m-[\text{O}-(\text{CH}_2)_n]_p-\text{O}-(\text{C}_1\text{-C}_{12} \text{ alkyl})$ ;

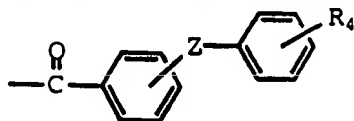
m and n are independently 2, 3 or 4;

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p is 0 or 1; or

C)  $R_3$  is  $-Y-(C_1-C_{12} \text{ alkyl})$ ;Y is  $-C\equiv C-$  or  $-CH=CH-$ ; orD)  $R_3$  is  $-O-(CH_2)_q-G$ ;

q is 2, 3 or 4;

G is  $C_7-C_{10}$  bicycloalkyl or  $C_7-C_{14}$  tricycloalkyl; orII)  $R_2$  is a group of the formula

where:

Z is  $-O-$ ,  $-C\equiv C-$ ,  $-CH=CH-$ ,  $-CH_2-CH_2-$ ,  $-CH_2-$ , or a bond;

A)  $R_4$  is hydrogen,  $C_1-C_{12}$  alkyl,  $C_1-C_{12}$  substituted alkyl,  $C_2-C_{12}$  alkenyl,  $C_2-C_{12}$  substituted alkenyl,  $C_2-C_{12}$  alkynyl,  $C_2-C_{12}$  substituted alkynyl,  $C_1-C_{12}$  alkoxy,  $C_3-C_{12}$  cycloalkyl,  $C_7-C_{10}$  bicycloalkyl,  $C_7-C_{14}$  tricycloalkyl,  $C_3-C_{12}$  cycloalkoxy, naphthyl, pyridyl, thienyl, benzothienyl, quinolyl or phenyl; or

B)  $R_4$  is phenyl substituted by amino,  $C_1-C_{12}$  alkylthio, halo,  $C_1-C_{12}$  alkyl,  $C_2-C_{12}$  alkenyl,  $C_2-C_{12}$  alkynyl,  $C_1-C_{12}$  substituted alkyl,  $C_2-C_{12}$  substituted alkenyl,  $C_2-C_{12}$  substituted alkynyl,  $C_1-C_{12}$  alkoxy, trifluoromethyl, phenyl, substituted phenyl, or phenyl substituted with a group of the formula  $-O-(CH_2)_m-[O-(CH_2)_n]_p-O-(C_1-C_{12} \text{ alkyl})$  where m, n and p are as defined above; or

C)  $R_4$  is  $C_1-C_{12}$  alkoxy substituted with halo,  $C_3-C_{12}$  cycloalkyl,  $C_7-C_{10}$  bicycloalkyl,  $C_7-C_{14}$  tricycloalkyl,  $C_1-C_6$  alkoxy,  $C_2-C_{12}$  alkynyl, amino,  $C_1-C_4$  alkylamino, di( $C_1-C_4$  alkyl)amino, formamido,  $C_2-C_{12}$  alkanoylamino, or phenyl substituted with a group of the formula  $-O-(CH_2)_m-[O-(CH_2)_n]_p-O-(C_1-C_{12} \text{ alkyl})$  where m, n and p are as defined above; or

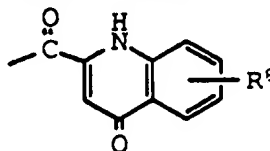
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D)  $R_4$  is  $-O-(CH_2)_r-W-R_5$ ;  
 $r$  is 2, 3 or 4;  
 $W$  is pyrrolidino, piperidino or piperazino;  
 $R_5$  is hydrogen,  $C_1$ - $C_{12}$  alkyl,  $C_3$ - $C_{12}$  cycloalkyl, benzyl or  $C_3$ - $C_{12}$  cycloalkylmethyl; or

E)  $R_4$  is  $-Y^1-R_6$ ;  
 $Y^1$  is  $-C\equiv C-$  or  $-CH=CH-$ ;  
 $R_6$  is  $C_3$ - $C_{12}$  cycloalkyl,  $C_7$ - $C_{10}$  bicycloalkyl,  $C_7$ - $C_{14}$  tricycloalkyl,  $C_3$ - $C_{12}$  cycloalkenyl, naphthyl, benzothiazolyl, thienyl, indanyl, fluorenyl, or phenyl substituted with  $C_1$ - $C_{12}$  alkylthio,  $C_2$ - $C_{12}$  alkenyl,  $C_2$ - $C_{12}$  alkynyl, halo( $C_1$ - $C_6$  alkoxy) or a group of the formula  $-O-(CH_2)_r-W-R_5$  where  $r$ ,  $W$  and  $R_5$  are as defined above; or  
 $R_6$  is phenyl substituted with a group of the formula  $-O-(CH_2)_m-[O-(CH_2)_n]_p-O-(C_1-C_{12} \text{ alkyl})$  where  $m$ ,  $n$  and  $p$  are as defined above; or

F)  $R_4$  is  $C_1$ - $C_{12}$  alkoxy substituted with a group of the formula  $-NHC(O)R_7$ ;  
 $R_7$  is  $C_1$ - $C_6$  alkoxy, or phenyl( $C_1$ - $C_6$  alkoxy); or

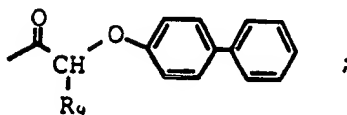
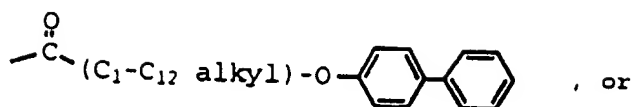
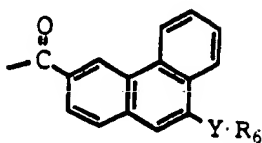
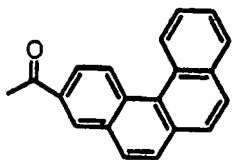
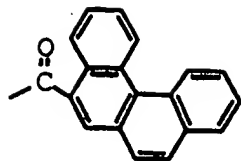
III)  $R_2$  is a group of the formula



where  $R^8$  is  $C_1$ - $C_{12}$  alkoxy or a group of the formula  $-O-(CH_2)_m-[O-(CH_2)_n]_p-O-(C_1-C_{12} \text{ alkyl})$  where  $m$ ,  $n$  and  $p$  are as defined above; or

IV)  $R_2$  is a group of the formula

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where:

Y and R<sub>6</sub> are as defined above;

R<sub>9</sub> is phenyl, C<sub>1</sub>-C<sub>12</sub> alkyl, or C<sub>1</sub>-C<sub>12</sub> alkoxy; or

V) R<sub>2</sub> is naphthoyl substituted with R<sub>4</sub> where R<sub>4</sub> is as defined above;

or a pharmaceutically acceptable salt thereof.

2. A compound according to claim 1 where:

R', R'', and R''' are each methyl;

RY<sup>1</sup>, RY<sup>2</sup>, RY<sup>3</sup>, RY<sup>4</sup> are each hydroxy;

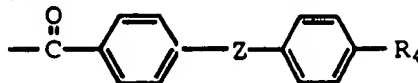
RX<sup>1</sup> and RX<sup>2</sup> are hydrogen;

or a pharmaceutically acceptable salt thereof.



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3. A compound according to claim 2 where:  
 $R_2$  is a group of the formula



where:

$Z$  is  $-\text{C}\equiv\text{C}-$  or a bond;

A)  $R_4$  is hydrogen,  $\text{C}_2\text{-C}_{12}$  alkynyl,  $\text{C}_2\text{-C}_{12}$  substituted alkynyl,  $\text{C}_1\text{-C}_{12}$  alkoxy,  $\text{C}_3\text{-C}_{12}$  cycloalkoxy, or phenyl; or

B)  $R_4$  is phenyl substituted by  $\text{C}_1\text{-C}_{12}$  alkoxy, or a group of the formula  $-\text{O}-(\text{CH}_2)_2-\text{O}-(\text{C}_1\text{-C}_6 \text{ alkyl})$ ; or

C)  $R_4$  is  $\text{C}_1\text{-C}_{12}$  alkoxy substituted with  $\text{C}_3\text{-C}_{12}$  cycloalkyl,  $\text{C}_7\text{-C}_{14}$  tricycloalkyl,  $\text{C}_1\text{-C}_6$  alkoxy, amino, di( $\text{C}_1\text{-C}_4$  alkyl)amino, formamido, or phenyl substituted with a group of the formula  $-\text{O}-(\text{CH}_2)_2-\text{O}-(\text{C}_1\text{-C}_6 \text{ alkyl})$ ; or

D)  $R_4$  is  $-\text{O}-(\text{CH}_2)_r-\text{W}-\text{R}_5$ ;  
 $r$  is 2 or 3;  
 $W$  is piperidino;  
 $R_5$  is hydrogen,  $\text{C}_1\text{-C}_{12}$  alkyl,  $\text{C}_3\text{-C}_{12}$  cycloalkyl, benzyl or  $\text{C}_3\text{-C}_{12}$  cycloalkylmethyl; or

E)  $R_4$  is  $-\text{Y}^1-\text{R}_6$ ;  
 $\text{Y}^1$  is  $-\text{C}\equiv\text{C}-$ ;  
 $R_6$  is phenyl substituted with  $\text{C}_1\text{-C}_{12}$  alkylthio,  $\text{C}_2\text{-C}_{12}$  alkenyl,  $\text{C}_2\text{-C}_{12}$  alkynyl, halo( $\text{C}_1\text{-C}_6$  alkoxy);  
or  $R_6$  is phenyl substituted with a group of the formula  $-\text{O}-(\text{CH}_2)_r-\text{W}-\text{R}_5$  where  $r$ ,  $W$  and  $R_5$  are as defined above;  
or  $R_6$  is phenyl substituted with a group of the formula  $-\text{O}-(\text{CH}_2)_2-\text{O}-(\text{C}_1\text{-C}_6 \text{ alkyl})$ ;

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F)  $R_4$  is  $C_1$ - $C_{12}$  alkoxy substituted with a group of the formula  $-NHC(O)R_7$ ;

$R_7$  is  $C_1$ - $C_6$  alkoxy, or phenyl ( $C_1$ - $C_6$  alkoxy); or a pharmaceutically acceptable salt thereof.

4. A compound according to claim 3 where  $R_1$  is  $C_1$ - $C_4$  alkyl,  $C_1$ - $C_4$  alkoxy, phenyl, p-chloro-phenyl, p-bromo-phenyl, or p-nitro-phenyl, benzyl, p-chloro-benzyl, p-bromo-benzyl, or p-nitro-benzyl; or a pharmaceutically acceptable salt thereof.

5. A compound according to claim 4 where  $R_1$  is methyl, ethyl, methoxy, ethoxy, phenyl, benzyl; or a pharmaceutically acceptable salt thereof.

6. The compound according to claim 5 where Z is a bond;  $R_4$  is phenyl substituted by n-pentoxy; and  $R_1$  is methyl; or a pharmaceutically acceptable salt thereof.

7. A pharmaceutical formulation comprising a compound of formula I or a pharmaceutically acceptable salt thereof, as claimed in any one of claims 1 to 6, associated with one or more pharmaceutically acceptable carriers, diluents or excipients therefor.

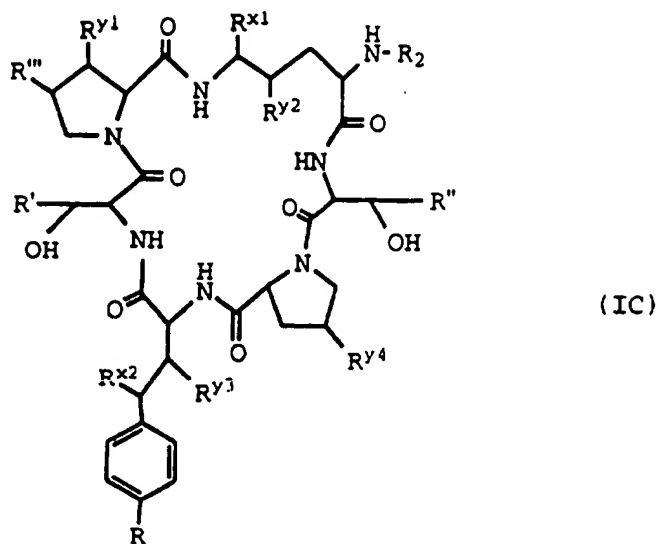
8. A compound of formula I, or a pharmaceutically acceptable salt thereof, as claimed in any of claims 1 to 6 for use as a pharmaceutical.

9. A method of treating a fungal infection in a mammal which comprises administering to such a mammal a compound of formula I, or a pharmaceutically acceptable salt thereof, as claimed in any one of claims 1 to 6.

10. A process for preparing a compound of formula I, or a pharmaceutically acceptable salt thereof as claimed in

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any of claims 1 to 6, comprising reacting a compound of formula IC



where R', R'', R''', Rx1, Rx2, Ry1, Ry2, Ry3, Ry4 and R<sub>2</sub> are as defined in claim 1;

with a substituted C<sub>1</sub>-C<sub>6</sub> alkyl phosphoric acid, phenyl phosphoric acid, substituted C<sub>1</sub>-C<sub>6</sub> alkyl phosphate or phenyl phosphate;

followed, if desired, by forming a pharmaceutically acceptable salt thereof.

## INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US96/04543

## A. CLASSIFICATION OF SUBJECT MATTER

IPC(6) : A61K 38/12; C07K 7/56

US CL : 514/11; 530/317

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 514/11; 530/317

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0561639 A1 (EIL LILLY AND COMPANY) 22 September 1993, see pages 2 and 34.	1-10

☐ Further documents are listed in the continuation of Box C. ☐ See patent family annex.

* "A"	Special categories of cited documents: document defining the general state of the art which is not considered to be of particular relevance	"T"	later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"E"	earlier document published on or after the international filing date	"X"	document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"L"	document which may throw doubt on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"Y"	document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
"O"	document referring to an oral disclosure, use, exhibition or other means	"Z"	document member of the same patent family
"P"	document published prior to the international filing date but later than the priority date claimed		

Date of the actual completion of the international search

08 JULY 1996

Date of mailing of the international search report

15.07.96

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